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## Changes of carbon dioxide in surface waters during spring in the Southern Ocean

D. C. E. BAKKER,\* H. J. W. DE BAAR\* and U. V. BATHMANN†

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**Abstract**—The fugacity of  $\text{CO}_2$  ( $f\text{CO}_2$ ) and the content of chlorophyll *a* in surface-water were determined during consecutive sections between  $47^\circ$  and  $60^\circ\text{S}$  along  $6^\circ\text{W}$  in austral spring, October–November 1992. In the Polar Frontal region, the  $f\text{CO}_2$  of surface-water decreased from slightly below the atmospheric value to  $50 \mu\text{atm}$  below it. This was accompanied by the development of diatom blooms. Seasonal warming of  $1.2^\circ\text{C}$  and air–sea exchange partly compensated the decrease of  $f\text{CO}_2$  by biological activity. Meanders of the Polar Frontal jet and a mesoscale eddy were reflected in spatial variability of  $f\text{CO}_2$  and chlorophyll *a*. Systematic observations indicated relationships between  $f\text{CO}_2$  and chlorophyll *a*, albeit changing with time. The combination of biological  $\text{CO}_2$ -uptake with formation of Antarctic Intermediate Water (AAIW) makes the Polar Front a site of combined biological/physical  $\text{CO}_2$ -drawdown from the atmosphere.

In the southern part of the Antarctic Circumpolar Current (sACC) and the Southern Frontal region,  $f\text{CO}_2$  increased  $7$ – $8 \mu\text{atm}$  due to surface-water warming of  $0.5^\circ\text{C}$ . A sharp rise of surface water  $f\text{CO}_2$  of  $13 \mu\text{atm}$  occurred south of the southern Frontal jet. As the ice-cover disappeared, the Boundary between the ACC and the Weddell Gyre released significant amounts of  $\text{CO}_2$ . The Weddell Gyre would become a strong  $\text{CO}_2$ -source after the imminent retreat of the ice. Clearly mechanisms behind changes of  $f\text{CO}_2$  in surface waters differ for the hydrographic regions. Interstitial brines of sea–ice had  $f\text{CO}_2$  as low as  $100 \mu\text{atm}$  and had been depleted in nutrients.

The summation of significant sources and sinks in the different regions indicates an overall minor oceanic  $\text{CO}_2$ -sink of  $0.3 \text{ mmol m}^{-2} \text{ day}^{-1}$  throughout the cruise, on the basis of the Wanninkhof relationship at *in situ* wind speed without skin effect. Uptake of  $\text{CO}_2$  increased to  $1.0 \text{ mmol m}^{-2} \text{ day}^{-1}$ , when a uniform cold skin temperature difference of  $0.2^\circ\text{C}$  was assumed. The skin temperature difference derived from the physical model by Soloviev and Schlüssel (1994a,b) had an average value of  $0.2^\circ\text{C}$ , leading to an uptake of  $\text{CO}_2$  of  $1.2 \text{ mmol m}^{-2} \text{ day}^{-1}$ . The measured skin temperature difference exceeded the calculated value. These assessments underline the uncertainty in the estimated air–sea exchange of  $\text{CO}_2$  due to the thermal skin effect, the chosen parametrization of the gas transfer velocity, and the selected length of the wind speed interval. Limited understanding of the mechanistics of gas exchange, as well as large seasonal and spatial variability of the air–sea flux, still preclude a reliable estimate of the basin-wide annual flux for the Southern Ocean. © 1997 Elsevier Science Ltd. All rights reserved

## INTRODUCTION

Increased atmospheric levels of carbon dioxide ( $\text{CO}_2$ ) account for about 60% of the  $\text{CO}_2$  emitted by fossil fuel burning and cement making. The remainder of these emissions is taken up by the oceans and terrestrial systems. Deforestation, fertilization by  $\text{CO}_2$ , and changes in land use and in carbon stocks in woods and soils complicate any estimate of the net uptake

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of fossil fuel CO<sub>2</sub> by terrestrial systems. The oceanic uptake of CO<sub>2</sub> is estimated as a range of 17–39% of the fossil fuel emissions (Sarmiento and Sundquist, 1992; Schimel *et al.*, 1995; Siegenthaler and Sarmiento, 1993; Tans *et al.*, 1990; among others).

The exchange of CO<sub>2</sub> between the oceans and the atmosphere is governed by physical transport at the air–water interface (Liss, 1983; Liss and Merlivat, 1986; Wanninkhof, 1992). The concentration difference across the interface drives the net flux  $F$  for gas exchange. For the purpose of this paper, this flux of CO<sub>2</sub> across the air–water interface can be expressed as:

$$F = k \cdot ([\text{CO}_2'(\text{aq})]_{\text{bulk}} - K_0 \cdot f\text{CO}_{2\text{air}}), \quad (1)$$

where  $k$  is the gas transfer velocity,  $[\text{CO}_2'(\text{aq})]_{\text{bulk}}$  ( $\mu\text{mol kg}^{-1}$ ) is the concentration of dissolved CO<sub>2</sub> in bulk surface-water,  $f\text{CO}_{2\text{air}}$  is the fugacity of CO<sub>2</sub> in marine air, and  $K_0$  is the solubility of CO<sub>2</sub> in seawater. The gas transfer velocity,  $k$  ( $\text{cm h}^{-1}$ ), is a function of interfacial turbulence, the kinematic viscosity  $\mu$  of the water, and the molecular diffusion coefficient  $D$  of the gas, both  $\mu$  and  $D$  in themselves depending on salinity and temperature. The fugacity of CO<sub>2</sub>,  $f\text{CO}_2$  ( $\mu\text{atm}$ ), is the partial pressure of CO<sub>2</sub>,  $p\text{CO}_2$ , corrected for its slightly non-ideal behaviour in the gas phase. Both the fugacity and the partial pressure can be calculated from the detected mixing ratio of CO<sub>2</sub> vs other gases in dry air (Wanninkhof and Thoning, 1993). Weiss (1974) provides a relationship for the solubility of CO<sub>2</sub> in water,  $K_0$  ( $\text{mol kg}^{-1} \text{atm}^{-1}$ ), as a function of temperature and salinity,

$$[\text{CO}_2'(\text{aq})] = K_0 \cdot f\text{CO}_2, \quad (2)$$

where  $[\text{CO}_2'(\text{aq})]$  includes a negligible amount of carbonic acid ( $\text{H}_2\text{CO}_3$ ). In practice, the term  $[\text{CO}_2'(\text{aq})]_{\text{bulk}}$  of above Equation (1) is also determined by relying on Equation (2) with measurement of  $f\text{CO}_2$  in the headspace of a seawater equilibrator.

Mixing ratios of atmospheric CO<sub>2</sub> in dry air are relatively uniform over large distances due to high lateral mixing rates. They are affected mainly by exchanges with the terrestrial biosphere and the oceans, anthropogenic emissions and point sources, such as volcanic eruptions. Concentrations of CO<sub>2</sub> in the oceanic mixed layer are influenced by temperature changes, biological activity, mixing of water masses, and upwelling. Dissolved CO<sub>2</sub> constitutes about 1% of the total dissolved inorganic carbon (DIC) in seawater. Other components are bicarbonate ( $\text{HCO}_3^-$ ) ( $\sim 90\%$ ), carbonate ( $\text{CO}_3^{2-}$ ) ( $\sim 9\%$ ) and carbonic acid ( $\text{H}_2\text{CO}_3$ ) ( $\sim 0.002\%$ ). Equilibration of CO<sub>2</sub> in surface-water with atmospheric CO<sub>2</sub> through air–sea exchange involves the complete carbonate system of seawater, and is therefore a relatively slow process in comparison with “non-reactive” gases such as H<sub>2</sub>, Ar and O<sub>2</sub> in seawater. In the open ocean, the  $f\text{CO}_2$  of surface-waters is generally within 15% of atmospheric equilibrium, though under- and oversaturations as much as 60% have been observed (Broecker *et al.*, 1986).

Major uncertainties exist when relating gas transfer velocities  $k$  to a function of wind speed. This is due to various sea state factors (wave height, swell, breaking waves, bubble entrainment, and others), whose individual dependencies on actual wind speed and wind history are not well quantified. These uncertainties become apparent when considering that two often used relationships by Liss and Merlivat (1986) and Wanninkhof (1992) yield fluxes that may differ by as much as 75%. Liss and Merlivat (1986) distinguished three wind speed regimes for the gas transfer velocity  $k_{\text{lm}}$  ( $\text{cm h}^{-1}$ ),

$$k_{\text{lm}} = 0.17 \cdot u \cdot (600/Sc)^{2/3} \quad u \leq 3.6 \quad (3a)$$

$$k_{lm} = (2.85 \cdot u - 9.65) \cdot (600/Sc)^{0.5} \quad 3.6 < u \leq 13 \quad (3b)$$

$$k_{lm} = (5.9 \cdot u - 49.3) \cdot (600/Sc)^{0.5} \quad u > 13, \quad (3c)$$

where  $Sc$  is the Schmidt number ( $Sc = \mu D^{-1}$ ), a dimensionless ratio of the transfers of momentum and mass (Liss and Merlivat, 1986; Wanninkhof, 1992), and  $u$  is the wind speed at 10 m above sea level ( $\text{m s}^{-1}$ ). The Liss–Merlivat relationship is based on measurements over 1–2 day intervals, and hence is more applicable to instantaneous rather than long-term flux assessment (Wanninkhof, 1992). Wanninkhof (1992) suggested two relationships for gas transfer velocities  $k_{w-sh}$  and  $k_{w-av}$  ( $\text{cm h}^{-1}$ ); one for steady winds, shipboard measurements as well as for scatterometer and radiometer data  $u_{sh}$  (Equation (4a)), and another for long-term averages of wind speed  $u_{av}$  (Equation (4b)).

$$k_{w-sh} = 0.31 \cdot u_{sh}^2 \cdot (660/Sc)^{0.5} \quad (4a)$$

$$k_{w-av} = 0.39 \cdot u_{av}^2 \cdot (660/Sc)^{0.5}. \quad (4b)$$

A temperature gradient may exist across the first millimeter of water at the sea surface (Robinson *et al.*, 1984). This thermal skin effect influences air–sea gas exchange mainly by changing the solubility  $K_0$  for  $\text{CO}_2$  in Equation (1) at the sea surface (Robinson *et al.*, 1984; Robertson and Watson, 1992). The effect of skin temperature on the gas transfer velocity  $k$  from its dependence on the Schmidt number (Equations (3a)–(3c), (4a) and (4b)) is negligible. Empirical formulae for air–sea gas exchange, like those of Liss and Merlivat (1986) and Wanninkhof (1992), have ignored the skin effect. Parametrizations based on gas transfer velocities of gases with a zero atmospheric concentration, like radon, have not been affected by the thermal skin effect. Gas transfer velocities from other gases, like natural and bomb  $^{14}\text{C}$ , have been slightly overestimated, here assuming a generally cold skin effect. However, this neglect of the thermal skin effect is only one cause of uncertainty in the relationships between wind speed and air–sea exchange. For example, the rather anomalous gas transfer velocity derived from bomb  $^{14}\text{C}$  depends strongly on the underlying  $^{14}\text{C}$  inventory model (Hesshaimer *et al.*, 1994).

Direct measurements of surface-water  $f\text{CO}_2$  are scarce for the Southern Ocean, which, south of the Polar Front accounts for roughly 11% of the global oceans (Hellmer and Bersch, 1985). Below an overview of  $f\text{CO}_2$  in surface-water of the Southern Ocean will be given travelling in an eastward direction with the Antarctic Circumpolar Current. The degree of saturation of the  $\text{CO}_2$ -content in water with respect to the atmospheric  $\text{CO}_2$ -level was defined as the difference of the fugacity of  $\text{CO}_2$  between water and air,  $\Delta f\text{CO}_2$ . In the case of published partial pressures,  $p\text{CO}_2$  is mentioned rather than  $f\text{CO}_2$ , the difference of about 0.7% (Weiss, 1974) being rather trivial.

### Atlantic sector

Takahashi *et al.* (1993) compiled data of the partial pressure of  $\text{CO}_2$  in surface-water in January through April in the years 1984–1990 for the Atlantic sector of the Southern Ocean. The general picture is one of low  $\Delta p\text{CO}_2$  ranging from 0 to  $-100 \mu\text{atm}$  in the western Weddell Sea. Undersaturation also was indicated for surface water of the Weddell Sea for January 1973 (Takahashi and Chipman, 1982), June–July 1992 and December–January 1992–1993 (Hoppema *et al.*, 1995). Lowest  $p\text{CO}_2$ -values were obtained in austral summer.

### *Indian sector*

Along 20–30°E, the CO<sub>2</sub>-content of surface-water was supersaturated with respect to the atmospheric CO<sub>2</sub>-level from 46 to 58°S, while it fluctuated between over- and undersaturation from 58 to 65°S in January–February 1987 (Metzl *et al.*, 1991). For a similar longitudinal range surface water fCO<sub>2</sub> was above the atmospheric value near the Polar Front and below it from the Polar Front to 65°S in February 1993 (Robertson and Watson, 1995). Surface water at 65°S had a slightly supersaturated CO<sub>2</sub>-content between 30 and 45°E in February–March 1993 (Robertson and Watson, 1995).

From 60 to 90°E, surface-water had an undersaturated CO<sub>2</sub>-content, with a weak seasonal variation in the Polar Frontal region in January, February and August 1991 (Poisson *et al.*, 1993), while it was slightly supersaturated in July 1984 (Goyet *et al.*, 1991). From the Polar Front to 67°S the surface water CO<sub>2</sub>-content fluctuated between undersaturation and oversaturation in January through May in the years 1991–1993 (Metzl *et al.*, 1995; Poisson *et al.*, 1993). The CO<sub>2</sub>-content of surface water was undersaturated between 45°S, 60°E and 65°S, 80°E and from 45°E to 80°E along 65°S in February–March 1993 (Robertson and Watson, 1995).

Along 115 and 150°E, the pCO<sub>2</sub> of surface water increased in southward direction, with undersaturation of CO<sub>2</sub> between 45° and 51°S, values close to the atmospheric CO<sub>2</sub>-content between 51° and 61°S and supersaturation between 61° and 65°S in December–January 1983–1984 (Inoue and Sugimura, 1986, 1988). The CO<sub>2</sub>-content of surface water was both undersaturated and oversaturated along 65°S from 115 to 150°E in the same period.

### *Pacific sector*

Surface-water in the area between 50 and 60°S, 170° and 100°W was slightly undersaturated in CO<sub>2</sub> in February through May in the years 1984–1989 (Murphy *et al.*, 1991a,b). Undersaturated CO<sub>2</sub>-values were found in the Bellingshausen Sea from 59 to 68°S along 88°W in November–December 1992 (Bellerby *et al.*, 1995; Robertson and Watson, 1995).

### *Summary of Southern Ocean observations*

The above suggests a mosaic of fCO<sub>2</sub>-values in surface-water with both oversaturation and undersaturation in austral summer. Undersaturation of CO<sub>2</sub> in surface-waters appears rather general in the Atlantic and Pacific sectors, while the Indian sector shows large spatial and probably temporal variation. From these data it is not clear whether the Southern Ocean constitutes a net sink or source of CO<sub>2</sub> in summer, not to mention that in winter which has hardly been sampled. Mechanisms behind the observed fCO<sub>2</sub>-values in surface-water often were not identified. Minima of fCO<sub>2</sub> frequently did not correlate to elevated chlorophyll *a* levels (Robertson and Watson, 1995), i.e. biological effects were not easily identifiable. Spatial variability of fCO<sub>2</sub> at scales of 10–100 km was ascribed to local primary production as well as to mesoscale dynamic processes reflecting bottom topography (Poisson *et al.*, 1993). An improved understanding of the air–sea exchange of CO<sub>2</sub> and its mechanisms is necessary for the Southern Ocean. Observations of seasonal evolution may improve our understanding of the underlying processes and of their variability in time and space.

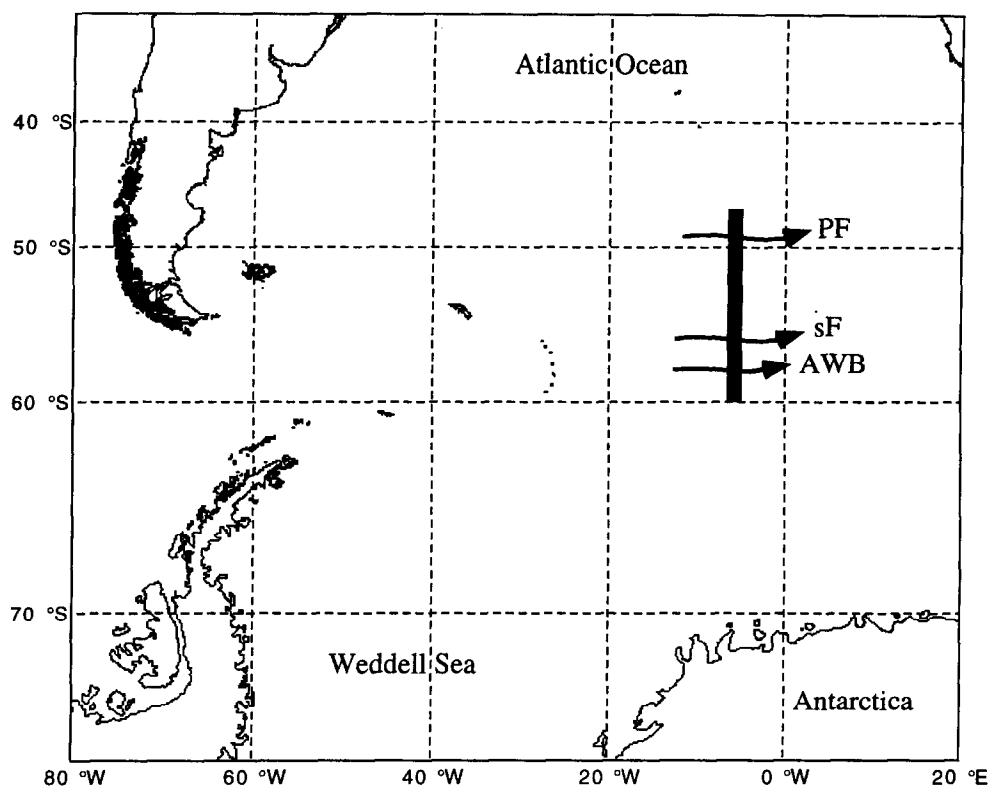


Fig. 1. The research area between 46.8° and 59.8°S along 6°W (thick line). The frontal jets of the Polar Front, the southern Front and the ACC-Weddell Gyre Boundary are indicated by black arrows.

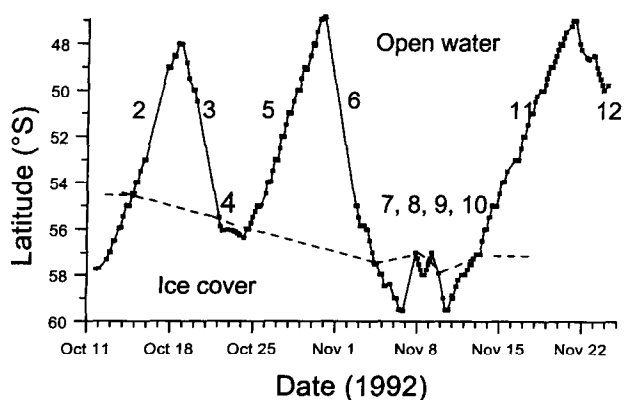


Fig. 2. The position of transects 2–12 and stations (squares) of ANT X/6 along 6°W between 11 October and 25 November 1992. The dashed line roughly indicates the position of the ice edge.

Measurements of  $f\text{CO}_2$  and of chlorophyll *a* in surface water were made at consecutive sections between 47 and 60°S along 6°W from 11 October to 24 November 1992 (Figs 1 and 2). The study focuses on the mechanisms that influence  $f\text{CO}_2$  in surface water. Air–sea exchange of  $\text{CO}_2$  is calculated with the Liss and Merlivat (1986) and Wanninkhof (1992) relationships. The effects of incorporation of a thermal skin effect and of the length of the wind speed interval on the estimated air–sea exchange fluxes are investigated.

## METHODS

Research onboard the German R.V. *Polarstern* was conducted during leg ANT X/6 of JGOFS Southern Ocean from 29 September to 29 November 1992 (Figs 1 and 2) (Bathmann *et al.*, 1994). Shipboard meteorological parameters were determined at the crow's nest at 30 m above sea level. The online data-acquisition system yielded 10 min averages of water temperature and salinity at 8 m depth, air temperature, atmospheric moisture content, atmospheric pressure at sea level, wind speed at 10 m above sea level, and wind direction. Salinity values were at the Practical Salinity Scale. The European Centre for Medium-Range Weather Forecasting (ECMWF) provided interpolated 3-hourly wind speed values for every half of a degree latitude between 47 and 60°S along 6°W. The ECMWF values were largely based on *Polarstern's* meteorological observations. Latitudinal average wind speeds from the ECMWF data varied between 10.6 and 11.0  $\text{m s}^{-1}$  for 11 October–24 November, and the regional average wind speed was 10.7  $\text{m s}^{-1}$ .

The fugacity of  $\text{CO}_2$  of water and marine air was measured semi-continuously with a gas chromatograph (GC), custom-built (Chrompack) by adaptation of the design of A.J. Watson (Plymouth Marine Laboratory). The GC contained two sampling loops of 0.8  $\text{cm}^3$ , one with an open outlet for the calibration gases and marine air, and a second as part of a closed system with the equilibrator headspace. Samples were not dried. During sampling the contents of one loop were brought into the GC's first column at atmospheric pressure under no-flow conditions. Two serial Haysep D columns of 0.5 and 1.0 m length, 2.1 mm inner diameter and a mesh width of 80–100  $\mu\text{m}$  allowed gases other than  $\text{CO}_2$  to be separated and vented. A nickel catalyst converted  $\text{CO}_2$  to methane ( $\text{CH}_4$ ), which was detected by a Flame Ionisation Detector (FID). Each 19 min run contained two calibration gases and subsequent samples for seawater, marine air, and seawater. Calibration gases supplied by BOC (U.K.) had mixing ratios of  $\text{CO}_2$  of 261.14, 361.27 and 476.45  $\mu\text{mol mol}^{-1}$  in artificial dry air, and had been calibrated against NOAA-certified standard gas mixtures.

Seawater was pumped from 12 m below sea level at a rate of 500  $\text{cm}^3 \text{s}^{-1}$  with a centrifugal pump. Sea-ice easily blocked the pump. A showerhead sprayed the seawater at a rate of 40–50  $\text{cm}^3 \text{s}^{-1}$  into the equilibrator, which had been constructed at NIOZ after a modification of the design of Watson (Fig. 3). Equilibrium between  $\text{CO}_2$  in the water and in the headspace takes place in less than 2 min. Atmospheric pressure in the headspace was maintained by a narrow bleeding valve. The headspace was connected to a closed system that contained a Capex L2 Charles Austen pump, a flowmeter, a 7  $\mu\text{m}$  dust filter, and a sampling loop of the GC. Gas was pumped around the closed system at 2  $\text{cm}^3 \text{s}^{-1}$  approximately half of the time between samplings to ensure flushing. One hundred percent moisture content was assumed in the headspace. Warming of the water in the equilibrator was minimised by maximal flow and insulation of tubing and equilibrator. Water temperature in the equilibrator was measured with a calibrated Pt-100 resistance thermometer. The average temperature difference between bulk water and that in the

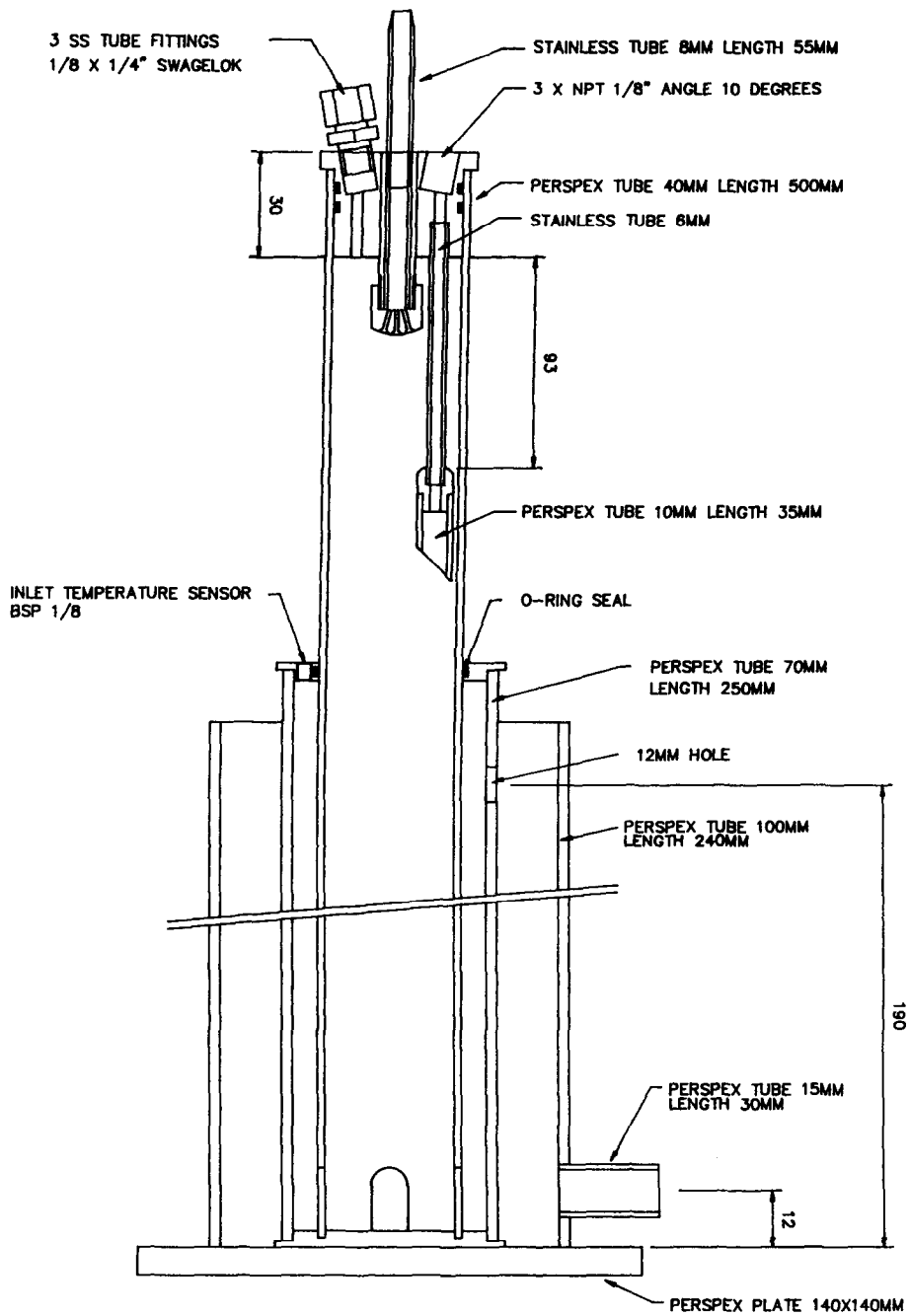


Fig. 3. The NIOZ equilibrator adapted after the design of A. J. Watson (Plymouth Marine Laboratory).



equilibrator was  $0.7^{\circ}\text{C}$ , but increased to  $1.5^{\circ}\text{C}$  near the ice. The equilibrator was placed under a black insulating cover. The fugacity of  $\text{CO}_2$  in bulk water was calculated from the mixing ratio of  $\text{CO}_2$ , atmospheric pressure, the Weiss (1974) formulae, and the temperature correction of Copin-Montégut (1988, 1989). Four successive samples, the last taken 1 h after the first, had an average standard deviation of  $0.7\ \mu\text{atm}$  for all samples of the transects 2, 3, 5, 6, 11 in the sACC and sF. The  $f\text{CO}_2$  of surface-water on all transects showed more variation in the Polar Frontal region due to biological blooms. The precision of  $f\text{CO}_2$  in surface-water was estimated as  $0.7\ \mu\text{atm}$  or  $0.2\%$ .

Marine air was pumped continuously from 20 m above sea level at a rate of  $80\ \text{cm}^3\ \text{s}^{-1}$  through 60 m long 0.25 in Dekabon tubing to the laboratory. Tubing was flushed before sampling. The mixing ratio of  $\text{CO}_2$  in dry air was calculated from the detected fraction in moist air and the ship's record of the atmospheric moisture content. A maximum intrinsic error of 10% in the atmospheric humidity amounts to an error of  $0.4\ \mu\text{mol mol}^{-1}$  or  $0.1\%$  in the mixing ratio of  $\text{CO}_2$  in dry air. Within the transects the standard deviation of the mixing ratios varied from 0.3 to  $0.6\ \mu\text{mol mol}^{-1}$ . The overall precision of the mixing ratios was estimated as  $0.6\ \mu\text{mol mol}^{-1}$  or  $0.2\%$ . Accuracy was estimated to be better than  $0.5\%$  from the mean offset between values from R.V. *Polarstern* and Syowa station at  $69^{\circ}00'\text{S}$ ,  $39^{\circ}35'\text{E}$  (Conway *et al.*, 1994), the residual difference being ascribed to a combination of regional and analytical variability. For flux calculations 100% atmospheric moisture content was assumed at the sea surface.

The  $f\text{CO}_2$  in marine air and surface-water and chlorophyll *a* is shown for transects 3, 5, 6 and 11 (Fig. 2). Transects 3 and 6 were southbound, and 5 and 11 were northbound. Time differences between transects for individual latitudes vary accordingly. The later transects (6, 11) extended further south because of the retreat of the ice-cover. Few measurements were made within the ice due to blockage of the pump.

Samples for oxygen ( $\text{O}_2$ ), total dissolved inorganic carbon (DIC) and alkalinity were taken from the CTD-rosette at standard depths at half and full degrees latitude as described in Bathmann *et al.* (1994). Oxygen samples were taken and analysed as duplicates with the Winkler method and a high precision oxygen titrator, following the recommendations of the WOCE Hydrographic Program (Culbertson, 1991). For calibration all chemicals including a stock solution of  $\text{KIO}_3$  had been preweighed. Oxygen measurements were carried out at temperatures below  $10^{\circ}\text{C}$  to prevent bubble formation. Detected values for duplicates were within  $0.4\ \mu\text{mol mol}^{-1}$  (Bathmann *et al.*, 1994).

DIC samples were poisoned with  $0.05\text{--}0.1\ \text{cm}^3$  saturated mercury(II)chloride solution to a content of  $20\text{--}50\ \mu\text{M HgCl}_2$ . For detection a subsample was acidified with 8.5% phosphoric acid to  $1\ \text{M H}_3\text{PO}_4$  and bubbled through with nitrogen. Gaseous  $\text{CO}_2$  was captured in an ethanol-amine solution with an indicator. The solution was photometrically backtitrated by a coulometer (Johnson *et al.*, 1987). Regular calibration was performed against DIC standard seawater samples supplied by A. G. Dickson (DOE, 1994). Precision amounted to  $1.5\ \mu\text{mol kg}^{-1}$  (Stoll, 1994).

Alkalinity samples of  $500\ \text{cm}^3$  were titrated with  $0.1\ \text{M}$  hydrochloric acid in a closed cell at  $20^{\circ}\text{C}$  after Bradshaw and Brewer (1988). A Gran plot was made of the datapoints after the second equivalence point. Carbon dissociation constants of Goyet and Poisson (1989) were used. The accuracy was about  $1\ \mu\text{eq kg}^{-1}$  (Stoll, 1994). Alkalinity–salinity distribution plots compared very well with those of nearby Geosecs stations 85–90.

The fluorescence of water obtained at 8 m depth with a membrane pump was registered with a Turner-Design TD 10 fluorometer in through-flow mode. Calibration was performed

at every station against fluorometric measurements of discrete water samples of usually  $1 \text{ dm}^3$  taken from a bypass next to the fluorometer. These samples were filtered through a GF/F glass fibre filter, homogenized for 5 min, and extracted in 90% acetone in the cold. Three times during the cruise both fluorometers were calibrated against SIGMA chlorophyll *a* standard from spinach extract analysed at a Shimadzu spectrophotometer. Fluorometer calibrations were stable (Bathmann *et al.*, 1997).

Skin temperature was detected and estimated from models. The skin temperature difference,  $\Delta T$ , was defined as the temperature difference between bulk water at 8 m depth and the skin layer. Thus, a positive skin temperature difference indicated a cool skin layer and an upward directed vertical heat flux.

Skin temperature was detected between 27 October–25 November 1992, with a Heimann KT 15.82A pyrometer suitable for irradiance between 8 and  $14 \mu\text{m}$  and temperatures between  $-25^\circ$  and  $75^\circ\text{C}$ . During calibration the pyrometer was adjusted on a black copper pipe that rested in a waterbath with a Pt-100 temperature sensor. Thus, a blackbody of well known temperature was simulated for a range from  $-5^\circ$  to  $10^\circ\text{C}$ . No difference was observed between the laboratory calibrations at 27 October and 25 November. Outdoors, the sensor had been mounted 20 m above sea level at an angle of  $40^\circ$  with the vertical. Care was taken to prevent the sensor from catching foam in its view. A thin plastic foil covered the lens, which did not appear to influence the signal. Emissivity values of the sea surface were estimated as a function of wind speed and instrument angle (Masuda *et al.*, 1988). The emissivities for wind speeds higher than  $15 \text{ m s}^{-1}$  were assumed to be equal to those at  $15 \text{ m s}^{-1}$ . A correction was made for radiation reflected from the sea surface, which supposedly originated from clouds, aerosols and atmospheric moisture. The temperature of the reflected radiation was estimated as that of the cloud base. This temperature was derived from three hourly observations of the height of the cloud base in 12 classes and daily vertical temperature profiles. Only measurements accompanying water temperatures of higher than  $-1.6^\circ\text{C}$  were used to avoid accidental ice floes in the path of the sensor.

The expected skin temperature difference was estimated with the parametrizations by Saunders (1967), Hasse (1971), Soloviev and Schlüssel (1994a,b) and the night-time model by Schlüssel *et al.* (1990). The corrections for the latter model (Soloviev and Schlüssel, 1996) were applied except for the sign of coefficient  $a_1$ , which was taken as positive. The models predict the skin temperature difference as a function of the net vertical heat flux through the sea surface. The model by Hasse includes solar radiation. The other models only apply during the night, but values calculated for the daytime have been included. Models have been developed and validated in temperate regions, generally under low wind conditions. The only model to include wind speeds above  $10 \text{ m s}^{-1}$  is that by Soloviev and Schlüssel (1994a,b). Sensible and latent heat fluxes were estimated from the atmospheric moisture content and water and air temperatures. Shipboard-detected global radiation was used as an estimate for solar radiation. Constant  $\lambda$  of 8 was assumed for the Saunders model. Drag coefficients, heat flux coefficients, and evaporation coefficients were adopted from Smith (1988). Values for the Prandtl number, kinematic viscosity, thermal diffusivity, latent heat of evaporation, specific heat capacity and the air density were derived from Beer (1983). Thermal expansion coefficients were taken from Unesco (1987). The contribution of net downwelling and upwelling infrared radiance was neglected.

Air-sea exchange of  $\text{CO}_2$  was calculated at *in situ* atmospheric pressure and 10 min shipboard wind speed for the Liss–Merlivat (Equations (3a)–(3c)) and Wanninkhof (Equation (4a)) relationships. Calculations were repeated several times by changing one

parameter each time and keeping the others constant. Subsequently the flux was estimated at an average atmospheric pressure of 1002.0 hPa, average wind speed of  $10.7 \text{ m s}^{-1}$  with the Wanninkhof relationship (Equation (4b)) and an assumed uniform cold skin effect of  $0.2^\circ\text{C}$ , as well as skin effects derived from the physical models and actual skin temperature measurements. Fluxes for the hydrographic regions were estimated from average values for every half of a degree latitude using zero air–sea exchange for the ice-covered areas. Total fluxes were calculated for  $46.8^\circ\text{--}59.8^\circ\text{S}$  at  $6^\circ\text{W}$  from weighted values for the hydrographic regions. Data from transects 2–12 were included in the total fluxes for the cruise. Zonal averages of the southern Frontal region (sF) and the ACC–Weddell Gyre Boundary (AWB) represent increasing areas of open water due to the retreating ice edge. The zero air–sea exchange in the ice-covered Weddell Gyre is included in estimates for the research area.

Sea-ice brine was sampled from two 15 cm diameter bore holes. Bore holes were 50 m apart at the first site, and 1 m apart at the second. For  $\text{fCO}_2$ -samples,  $600 \text{ cm}^3$  glass bottles were mechanically lowered into the brine, and their screw cap was lifted and turned on again after the brine had filled the bottle. This procedure was repeated for DIC, alkalinity and salinity samples. Nutrients were collected with a plastic syringe afterwards. Consequently, silicate data are maximum values as the glass bottles may have “contaminated” the brine.

## RESULTS

### Hydrography

The 1400 km stretch between  $47^\circ$  and  $60^\circ\text{S}$  at  $6^\circ\text{W}$  is within the eastward flowing Antarctic Circumpolar Current (ACC) and the Weddell Gyre (WG). It contains three fronts: the Polar Front (PF), the southern Front of the ACC (sF) and the ACC–Weddell Gyre Boundary (AWB). The fronts constitute boundaries within the open waters of the ACC and between the ACC and the Weddell Gyre. The southern Front is considered to be an interruption in the sACC. Within the research area, five regions have been distinguished: the Polar Frontal region (PF), the southern ACC (sACC), the southern Frontal region (sF), the Boundary region (AWB) and the Weddell Gyre (WG) (Fig. 1 and Table 1). Veth *et al.* (1997) have described the hydrography of the region in detail; some major features are summarised here.

The ice edge was at  $54.6^\circ\text{S}$  when the ship reached  $6^\circ\text{W}$  (Table 2). The ice edge was probably still near its northernmost position of the past winter (Veth, personal communication, 1994). This is in accordance with a maximum ice extension of about  $53^\circ\text{S}$

Table 1. Hydrographic regions between  $46.8^\circ$  and  $59.8^\circ\text{S}$  at  $6^\circ\text{W}$  (Veth *et al.*, 1997)

Hydrographic region		Latitude ( $^\circ\text{S}$ )	Remarks
Polar Frontal region	PF	$(46.8) \leq \text{PF} < 50.8$	Jet at $48.8^\circ\text{--}49.0^\circ\text{S}$ , meanders $47^\circ\text{--}49^\circ\text{S}$ , eddy $49.0^\circ\text{--}50.5^\circ\text{S}$
Southern Antarctic Circumpolar Current	sACC	$50.8 \leq \text{sACC} < 54.3$	
Southern Front	sF	$54.3 \leq \text{sF} < 56.3$	Jet at $55.8^\circ\text{S}$ , meanders north of jet
ACC–Weddell Gyre Boundary	AWB	$56.3 \leq \text{AWB} < 58.8$	Jet at $58.0^\circ\text{S}$
Weddell Gyre	WG	$58.8 \leq \text{WG} \leq (59.8)$	

Table 2. The time interval, position of the ice edge at 6°W and atmospheric mixing ratios of CO<sub>2</sub> in dry air for the transects. Mixing ratios for Syowa, Palmer and Ascension stations of NOAA/CMDL have been included (Conway *et al.*, 1994)

Transect	Period 1992	Ice edge position (°S)	xCO <sub>2</sub> ANT X/6, 6°W ( $\mu\text{mol mol}^{-1}$ )	xCO <sub>2</sub> Syowa 69°00'S, 39°35'E ( $\mu\text{mol mol}^{-1}$ )	xCO <sub>2</sub> Palmer 64°55'S, 64°00'W ( $\mu\text{mol mol}^{-1}$ )	xCO <sub>2</sub> Ascension 7°55'S, 14°25'W ( $\mu\text{mol mol}^{-1}$ )
2	11/10–18/10	54.7	357.0	355.2	355.4	355.8
3 (4)	18/10–23/10	55.5	356.5	355.2	355.4	355.8
5 (4)	23/10–31/10	56.0	356.0	355.2	355.4	355.8
6	31/10–06/11	57.4	356.0	355.2	355.0	355.6
11	10/11–21/11	57.7	356.0	355.2	355.0	355.6
12	21/11–24/11	—	355.5	355.2	355.0	355.6

at 6°W (Hellmer and Bersch, 1985). The ice edge retreated during the cruise to 57.7°S (Van Franeker in Bathmann *et al.*, 1994; Table 2). Strong winds dispersed the ice at the edge of the continuous ice-cover. The ice edge as such was generally not well defined, and transitions of ice-cover from 0 to 100% occurred over short distances in all directions. Surface-water salinity showed a slight minimum at sites where most melting took place (not shown). This minimum had disappeared during the subsequent transect, probably by mixing of the water. Yet another salinity minimum occurred in the actual melting zone.

Antarctic Intermediate Water is formed at the Polar Front when surface-water from the south sinks below water from the north. The Polar Frontal jet with high eastward surface flow velocities was found between 48.8° and 49.0°S. The Polar Front separated clearly distinct water masses near the sea surface. The Polar Frontal region (PF) was highly dynamic, with meanders between 47.0° and 49.0°S and a rather persistent eddy structure between 49.0° and 50.5°S. The southern ACC was relatively homogeneous over large distances. The southern Frontal jet was close to the ice edge at 55.8°S during transects 3 and 5. The position of the frontal jet of the boundary between the ACC and the Weddell Gyre at 58.0°S was defined by the penetration of Upper Circumpolar Deep Water at a certain depth. Surface-water characteristics, like salinity and water temperature, indicated a continuation of the same water mass across the front.

The occurrence of 10 min shipboard wind speeds is depicted in Fig. 4. Storms were encountered frequently, with wind velocities as high as  $24 \text{ m s}^{-1}$ . Latitudinal average wind speeds from the ECMWF data varied between  $10.6$  and  $11.0 \text{ m s}^{-1}$  from 11 October to 24 November, with a regional average of  $10.7 \text{ m s}^{-1}$  for 47°–60°S at 6°W. Mixed layer depths were between 60 and 120 m. Water temperature increased between transects in areas of open water (Fig. 5).

### Atmospheric CO<sub>2</sub>

The mixing ratios of CO<sub>2</sub> in dry air decreased from  $357.0 \mu\text{mol mol}^{-1}$  for transect 2 to  $355.5 \mu\text{mol mol}^{-1}$  for transect 12 (Table 2). These mixing ratios were  $0.3$ – $1.8 \mu\text{mol mol}^{-1}$  higher than values at Syowa station (69°00'S, 39°35'E) and Palmer station (64°55'S, 64°00'W) for October and November 1992 (Conway *et al.*, 1994). This is consistent with, if not exceeding, the general trend of higher mixing ratios of CO<sub>2</sub> towards the north. Values at

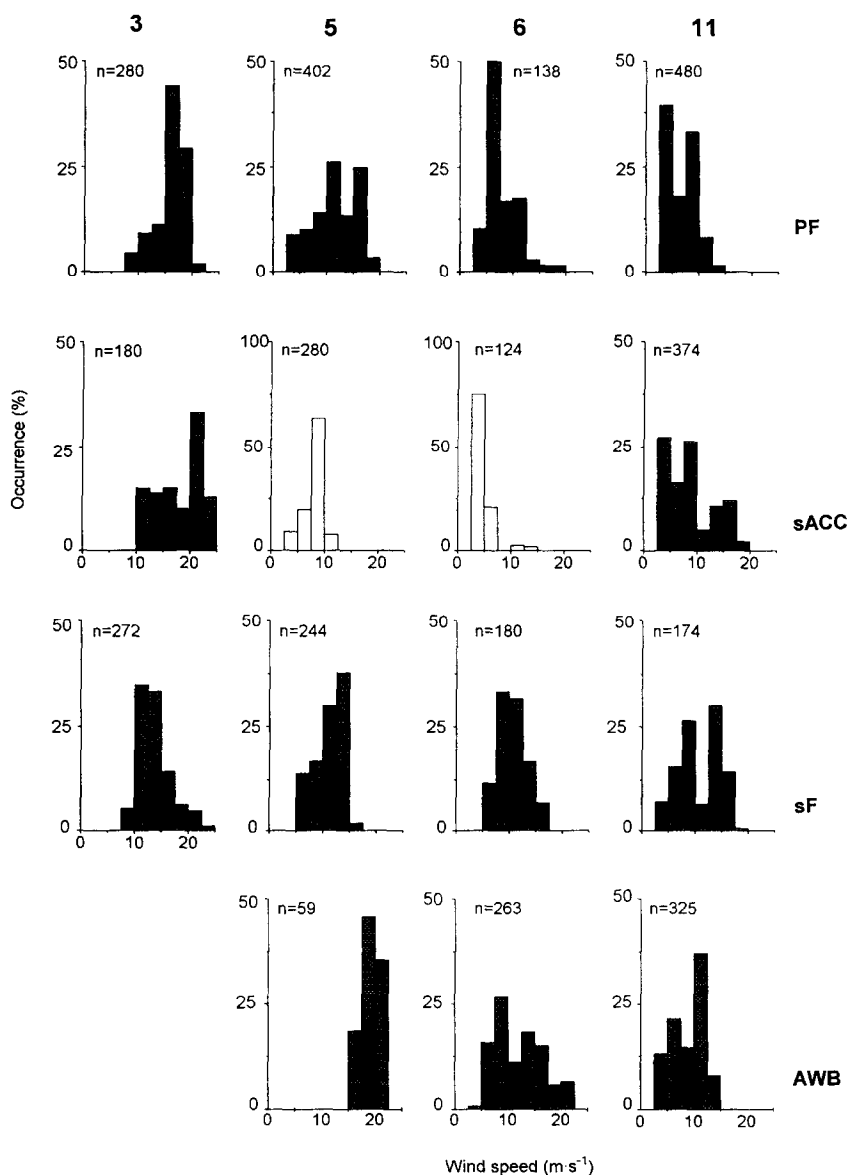


Fig. 4. Occurrence of 10 min shipboard wind speed values accompanying the  $f\text{CO}_2$  measurements for the hydrographic regions of transects 3, 5, 6 and 11 (Table 1). The number of datapoints is indicated in each figure. The thickness of the bars denotes a range of  $2.5 \text{ m s}^{-1}$ . The scale of vertical axis of graphs with grey bars is 50% occurrence, of those with white bars 100% occurrence.

Ascension Island ( $7^{\circ}55'S$ ,  $14^{\circ}25'W$ ) in the tropical South Atlantic Ocean were  $0.1$  higher to  $1.2 \mu\text{mol mol}^{-1}$  lower than our data. Over the years the shore-based stations, with intrinsically better accuracy than shipboard observations, show a likely natural, variation within  $\sim 1 \mu\text{mol mol}^{-1}$  for a given month (Amsterdam Island, Ascension Island, Cape Grim, Palmer Station, Syowa Station, all in Conway *et al.*, 1994).

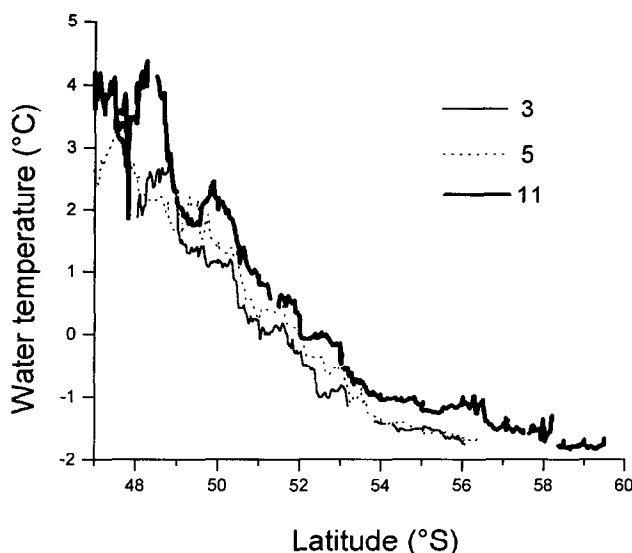


Fig. 5. Temperature of surface-water along 6°W for transects 3, 5 and 11 of ANT X/6 (Table 2).

Atmospheric pressure varied between 975 and 1024 hPa, with an average of 1002.0 hPa for transects 2–12. This affected the  $f\text{CO}_2$  air as indicated by its wave-like and variable appearance in Fig. 6. Changes in  $f\text{CO}_2$  air influence the concentration of dissolved  $\text{CO}_2$  at the sea surface, hence, the air–sea exchange. The effect on air–sea exchange is discussed below.

#### Surface-water $f\text{CO}_2$

**Polar Frontal region.** In the Polar Frontal region the  $\text{CO}_2$ -content of surface-water was undersaturated with respect to the atmospheric  $\text{CO}_2$ -content for transect 3 (Fig. 6). Chlorophyll *a* levels, albeit below  $1 \text{ mg m}^{-3}$ , appeared higher in the PF region and adjacent areas of the sACC than further south (Fig. 7). Surface-water  $f\text{CO}_2$  had decreased 10 days later during transect 5 (Fig. 6). This was accompanied by increases of primary production (Jochem *et al.*, 1995) and chlorophyll *a* (Fig. 7). Water was even more undersaturated with  $\text{CO}_2$ , and chlorophyll *a* levels had further increased 3 days later during transect 6. Elevated chlorophyll *a* levels extended to more than 100 m depth in the PF region, with a maximum of  $1.4 \text{ mg m}^{-3}$  at about 50 m depth at  $49.0^\circ\text{S}$  (Bathmann *et al.*, 1997). Large meanders of the Polar Front and the persistent eddy structure (Table 1; Veth *et al.*, 1997) were reflected in the variability of surface water  $f\text{CO}_2$  and the content of chlorophyll *a* (Figs 6 and 7).

Bloom development continued in the PF region during transects 11 and 12, correlating with a strong decrease of surface water  $f\text{CO}_2$  and  $[\text{CO}_2'(\text{aq})]$  (Figs 6–8). Meanders of the front and the eddy structure were reflected in abrupt changes of  $f\text{CO}_2$  and chlorophyll *a* in surface waters. Algal blooms with chlorophyll *a* maxima of  $3.5 \text{ mg m}^{-3}$  corresponded to  $f\text{CO}_2$  as low as  $\sim 300 \mu\text{atm}$ , while elsewhere surface water had relatively low chlorophyll *a* values of  $0.8 \text{ mg m}^{-3}$  and  $f\text{CO}_2$  of  $340\text{--}345 \mu\text{atm}$ . Algal blooms also appeared in contour plots of CTD-samples as three pronounced maxima of chlorophyll *a* of up to  $2.4 \text{ mg m}^{-3}$  at about 50 m depth, two in meanders of the front and one in the mesoscale eddy (Bathmann *et al.*, 1997).

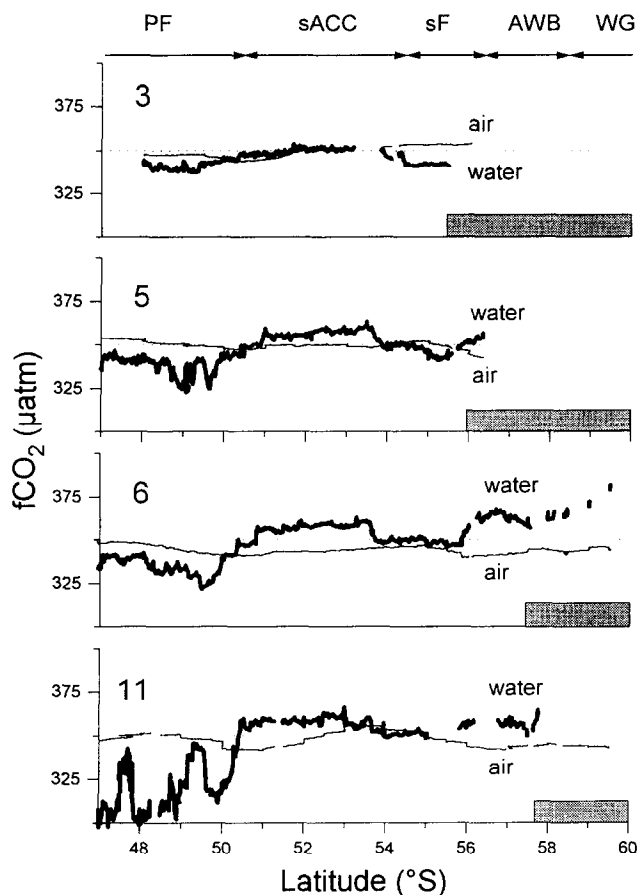


Fig. 6. The fugacity of  $\text{CO}_2$  ( $f\text{CO}_2$ ) in surface-water (thick line) and marine air (thin line) without a thermal skin effect for transects 3, 5, 6 and 11 of ANT X/6 along  $6^\circ\text{W}$  (Table 2). Shaded bars denote ice-cover. Hydrographic regions are indicated at the top of the graph.

Alkalinity in the upper 200 m water column did not change significantly throughout the cruise. This is consistent with the planktonic species composition, which did not indicate any calcifying activity (Scharek, personal communication; Bathmann *et al.*, 1997). DIC decreased with  $10\text{--}15\ \mu\text{mol kg}^{-1}$  in parts of the PF region between transects 5 and 11. The combination of  $f\text{CO}_2$  and DIC decreased, chlorophyll *a* increased, constant alkalinity, and absence of calcifying organisms suggested a net uptake of  $\text{CO}_2$  due to photosynthetic activity. The increase of water temperature throughout the cruise (Fig. 5) counteracted the effect of biology on  $f\text{CO}_2$ . In other words, the photosynthetic decrease of  $f\text{CO}_2$  was partly compensated by seasonal warming.

*Southern part of the Antarctic circumpolar current.* The  $\text{CO}_2$ -content of surface waters was near saturation with respect to the atmosphere during transect 3 in the southern part of the Antarctic Circumpolar Current (sACC) (Fig. 6). Between transects 3 and 11,  $f\text{CO}_2$ -values

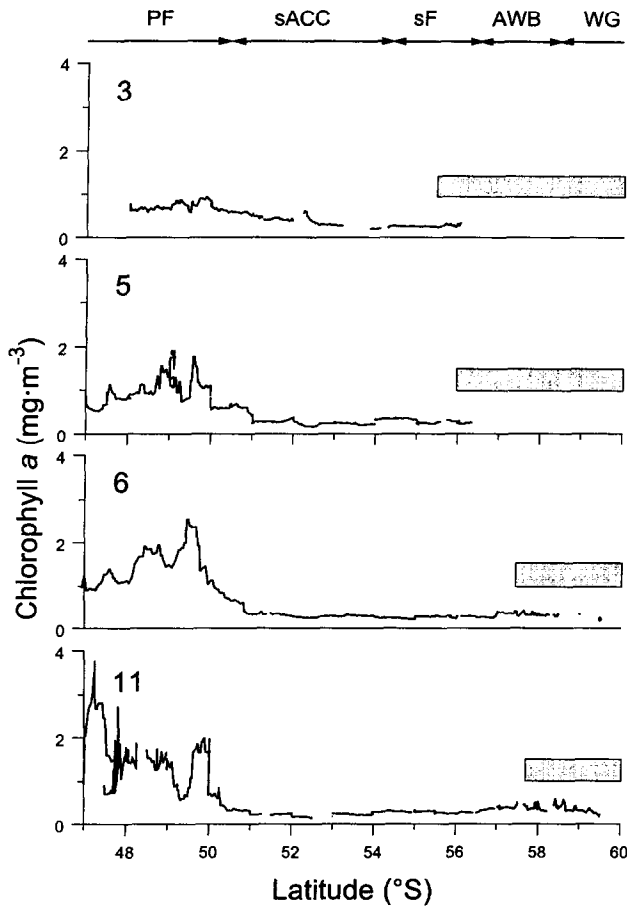


Fig. 7. The content of chlorophyll *a* in surface-water for transects 3, 5, 6 and 11 of ANT X/6 along 6°W (Table 2). Shaded bars denote ice-cover. Hydrographic regions are indicated at the top of the graph.

increased with  $8 \mu\text{atm}$  (Fig. 6), while  $[\text{CO}_2'(\text{aq})]$  remained virtually constant (Fig. 8). The increased surface water temperature with time (Fig. 5) would correspond to a thermodynamical increase of  $f\text{CO}_2$ , similar in size to the change observed, but would hardly change  $[\text{CO}_2'(\text{aq})]$ . Biological activity in the sACC was low throughout the cruise (Fig. 7).

To summarise, the observed increase of  $f\text{CO}_2$  in surface-waters of the sACC was largely the result of seasonal warming. It is conceivable that upwelling brings deeper waters with high  $\text{CO}_2$ -contents to the surface and thus contributes to an oversaturation of the  $\text{CO}_2$ -content of surface waters, but no such effect of upwelling was discerned in the  $\text{CO}_2$ -signal of the sACC.

*Southern Frontal region.* During transect 3, surface-water was undersaturated with  $\text{CO}_2$  between  $54.5^\circ$  and  $55.5^\circ\text{S}$  shortly after and during the breakup of the ice-cover (Fig. 6). The undersaturation could be ascribed to rapid cooling of the water in the preceding autumn and



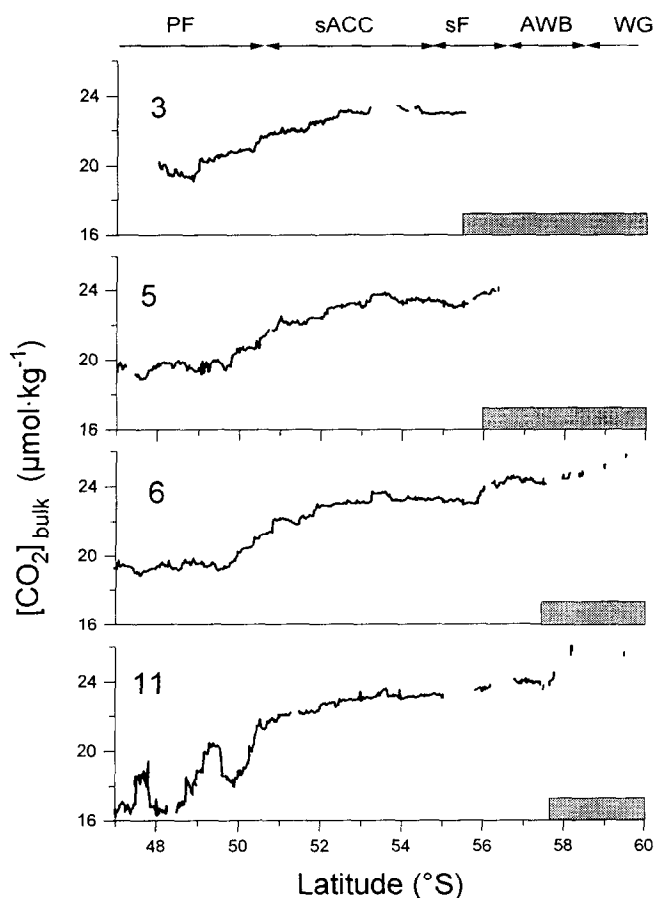


Fig. 8. The concentration of dissolved  $\text{CO}_2$  in surface-water for transects 3, 5, 6 and 11 of ANT X/6 along  $6^\circ\text{W}$  (Table 2). Shaded bars denote ice-cover. Hydrographic regions are indicated at the top of the graph.

subsequent ice-cover. This assumes that the cooling had taken place so quickly that replenishment of  $\text{CO}_2$  by air-sea exchange was inadequate to overcome the undersaturation. Both the  $f\text{CO}_2$  and the temperature of surface-waters increased between the transects, while  $[\text{CO}_2(\text{aq})]$  remained nearly constant (Figs 5, 6 and 8). Immediately south of the frontal jet, the  $f\text{CO}_2$  of surface-waters showed a marked increase of  $13 \mu\text{atm}$  during transect 6. South of the jet, surface waters were strongly supersaturated with  $\text{CO}_2$ . Chlorophyll *a* levels were low throughout this region and not obviously related to the frontal jet or the retreating ice edge (Fig. 7).

*ACC–Weddell Gyre Boundary region.* In the Boundary region between the ACC and the Weddell Sea, the  $\text{CO}_2$ -content of surface waters was strongly oversaturated during transects 6 and 11 (Fig. 6). Oversaturation was weakest at the position of the frontal jet at  $58.0^\circ\text{S}$ . Water south of the frontal jet had a variable, strongly supersaturated  $\text{CO}_2$ -content. Chlorophyll *a* levels were low during transects 6 and 11 (Fig. 7). The surface water between

the Boundary and the southern Front had been influenced by similar processes as water in the Weddell Gyre because the Boundary is a subsurface phenomenon and does not separate distinct surface water masses. There was no detectable effect of the melting on water temperature, nutrients,  $f\text{CO}_2$ , alkalinity and DIC.

*Weddell Gyre.* Water of the Weddell Gyre contained a highly variable, strongly supersaturated  $\text{CO}_2$ -content below the ice during transect 6 (Fig. 6). Oxygen saturation was 80–90% with an AOU (apparent oxygen utilization) of  $20\text{--}60\ \mu\text{mol kg}^{-1}$  during transects 6 and 11 (Fig. 9) (Bathmann *et al.*, 1994). Imminent retreat of the ice edge would cause considerable outgassing of  $\text{CO}_2$  to the atmosphere, most notably with respect to the generally high wind stress enhancing air–sea exchange.

#### *Surface-water $f\text{CO}_2$ vs chlorophyll *a**

During transect 3 surface water  $\text{CO}_2$ -contents close to equilibrium with the atmospheric  $\text{CO}_2$  content corresponded to chlorophyll *a* contents lower than  $1\ \text{mg m}^{-3}$  (Fig. 10). During later transects, two groups of data could be discerned. One group corresponded originally to surface-waters of the sACC and sF. Here chlorophyll *a* levels remained low, while  $f\text{CO}_2$  increased with time by surface water warming. High  $f\text{CO}_2$ -values with low chlorophyll *a* contents were added to this group as the transects extended further south into recently ice-free or still ice-covered regions of the AWB and the WG.

Elevated chlorophyll *a* levels of the second group were accompanied by undersaturated

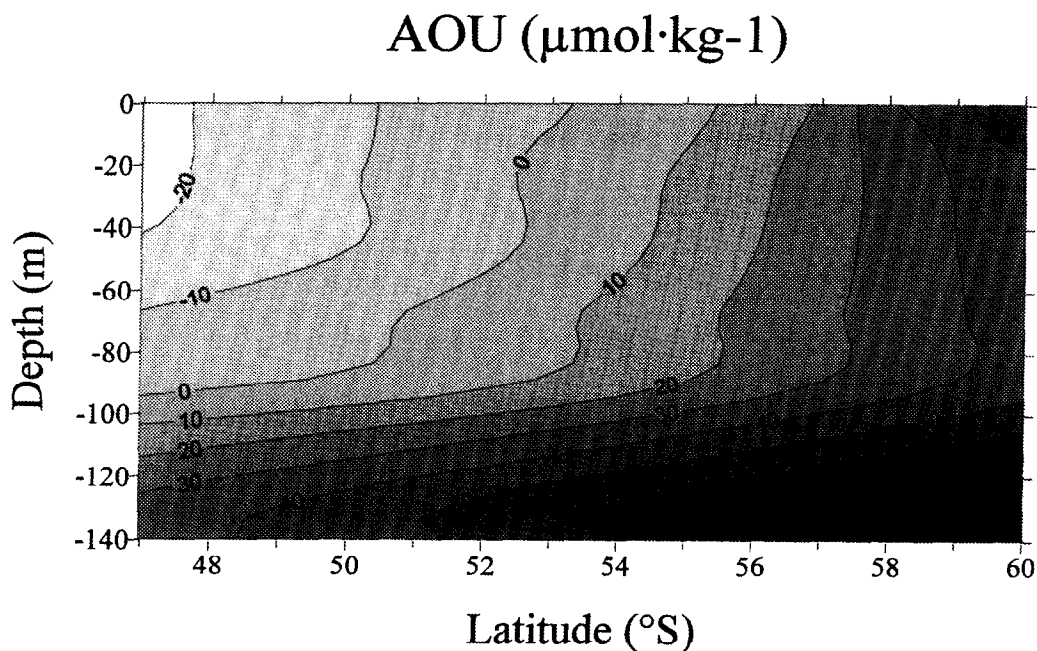


Fig. 9. The apparent oxygen utilization (AOU) ( $\mu\text{mol kg}^{-1}$ ) for the upper 140 m of water for transect 11 (Table 2). Negative values indicate supersaturation of dissolved  $\text{O}_2$  relative to the atmospheric value.

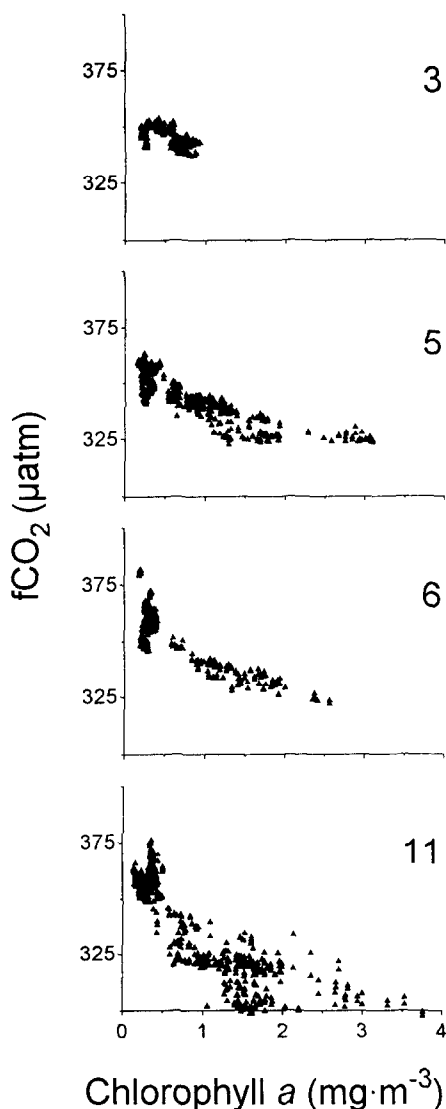


Fig. 10. The fugacity of  $\text{CO}_2$  ( $f\text{CO}_2$ ) as a function of the content of chlorophyll  $a$  in surface-water for transects 3, 5, 6 and 11 of ANT X/6 along  $6^\circ\text{W}$  (Table 2).

$\text{CO}_2$ -values. Chlorophyll  $a$  contents of 1 and  $3 \text{ mg m}^{-3}$  corresponded to  $f\text{CO}_2$  of 340 and  $325 \mu\text{atm}$ , respectively, during transects 5 and 6. The same contents of chlorophyll  $a$  were related to lower and more variable  $f\text{CO}_2$ -values of 320–340 and  $305 \mu\text{atm}$  during transect 11. This change was caused by locally different intensities of photosynthetic uptake of  $\text{CO}_2$  and seasonal warming. Air–sea exchange of  $\text{CO}_2$  tended to minimize the undersaturation or oversaturation.

### Air-sea exchange of $\text{CO}_2$

Fluxes at average atmospheric pressure were close to those at *in situ* atmospheric pressure (Table 3). Fluxes mentioned below have been calculated with the appropriate Wanninkhof relationship (Equation (4a) or (4b)) at *in situ* atmospheric pressure and initially without thermal skin effect.

In the Polar Frontal region, undersaturation of  $\text{CO}_2$  increased with time. The air-sea exchange calculated at average wind speed ( $10.7 \text{ m s}^{-1}$ ) would increase accordingly (Fig. 11, Table 4). The air-sea exchange flux at *in situ* wind speed did not show this consistent pattern due to wind speeds higher than average during transects 3 and 5 and lower than average during transects 6 and 11 (Figs 4 and 11). The latter low wind velocities are a prerequisite for bloom development under suitable light conditions in a stabilizing mixed layer (de Baar *et al.*, 1995). Thus, wind speed has opposing effects on the terms  $[\text{CO}_2'(\text{aq})]_{\text{bulk}}$  and  $k$  in the flux (Equation (1)). Photosynthetic  $\text{CO}_2$ -fixation leads to undersaturation of  $[\text{CO}_2'(\text{aq})]$ , but only intermittent or high wind events result in a transfer coefficient  $k$  large enough to produce a significant influx of  $\text{CO}_2$  from the atmosphere. Daily observations are therefore essential to determine a reliable long-term flux for the Polar Frontal region.

In the sACC, at an assumed average wind speed, initial negligible exchange of  $\text{CO}_2$  would evolve into an efflux of  $\text{CO}_2$  due to seasonal warming. The southern Frontal region turned from an oceanic sink into a source by seasonal warming. Here patterns for fluxes calculated at *in situ* and average wind speed were rather similar. Retreat of the ice edge uncovered surface waters supersaturated with  $\text{CO}_2$  and made the AWB an oceanic source of  $\text{CO}_2$ . High uptake of  $\text{CO}_2$  into the AWB at *in situ* wind speed during transect 6 corresponded to a storm with wind speeds exceeding  $20 \text{ m s}^{-1}$ . In the Weddell Gyre, ice-cover inhibited air-sea exchange.

The overall effect of the length of the wind speed interval on the exchange is shown in Fig. 12 (Table 3). Oceanic uptake of  $\text{CO}_2$  at average wind speed was considerably higher than at *in situ* wind speed for the research area in October–November 1992. At *in situ* wind speed,

Table 3. Sea to air exchange of  $\text{CO}_2$  ( $\text{mmol m}^{-2} \text{ day}^{-1}$ ) in the hydrographic regions (Table 1) along  $6^\circ\text{W}$  for ANT X/6 with the Liss and Merlivat (1986) and Wanninkhof (1992) relationships (equations (3a)–(4b)) at *in situ*, 10 min shipboard values for wind speed and atmospheric pressure without any thermal skin effect. Fluxes with the Wanninkhof relationship also have been estimated for average atmospheric pressure and wind speed without a skin effect. Fluxes at *in situ* wind speed and atmospheric pressure have been determined for a cold skin effect of  $0.2^\circ\text{C}$  and modelled skin effects by Hasse (1971) and Soloviev and Schlüssel (1994a, 1994b). Positive values indicate oceanic release of  $\text{CO}_2$ , negative fluxes denote oceanic uptake. See also Fig. 12

	PF	sACC	sacF	AWB	46.8°–59.8°S
<i>Liss–Merlivat</i>					
<i>In situ</i> values, $\Delta T = 0^\circ\text{C}$	–2.0	0.8	–0.2	1.5	–0.2
<i>Wanninkhof</i>					
<i>In situ</i> values, $\Delta T = 0^\circ\text{C}$	–3.7	1.4	–0.4	2.7	–0.3
Mean atm. pressure, $\Delta T = 0^\circ\text{C}$	–3.7	1.5	–0.3	2.8	–0.3
Mean wind, $\Delta T = 0^\circ\text{C}$	–6.4	1.9	–0.5	1.6	–1.2
$\Delta T = 0.2^\circ\text{C}$	–4.4	0.1	–1.1	2.3	–1.0
$\Delta T$ —Hasse	–3.8	1.3	–0.5	2.7	–0.4
$\Delta T$ —Soloviev and Schlüssel	–4.8	0.0	–0.9	2.2	–1.2

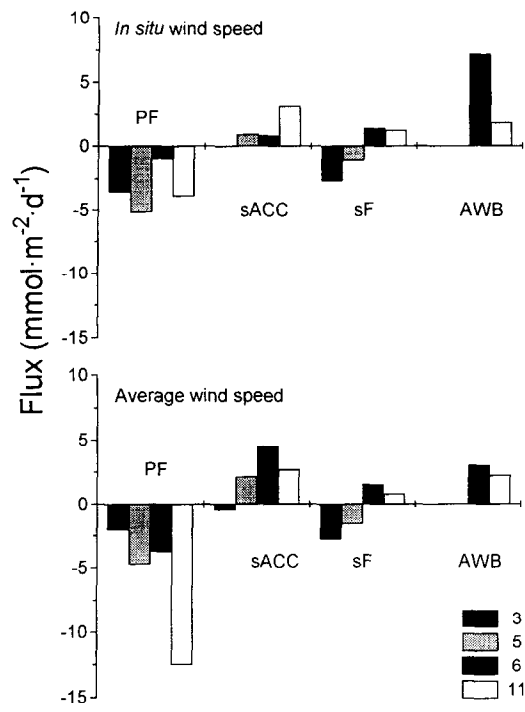


Fig. 11. Sea to air exchange of CO<sub>2</sub> for the hydrographic regions along 6°W (Table 1) for transects 3, 5, 6 and 11 of ANT X/6 at *in situ* (above) and average wind speed (below). Fluxes have been calculated with the appropriate Wanninkhof relationship (Equation (4a) and (4b), respectively) at *in situ* atmospheric pressure without a thermal skin effect. Note that positive fluxes indicate oceanic release of CO<sub>2</sub>, while negative values denote oceanic uptake. See also Table 4.

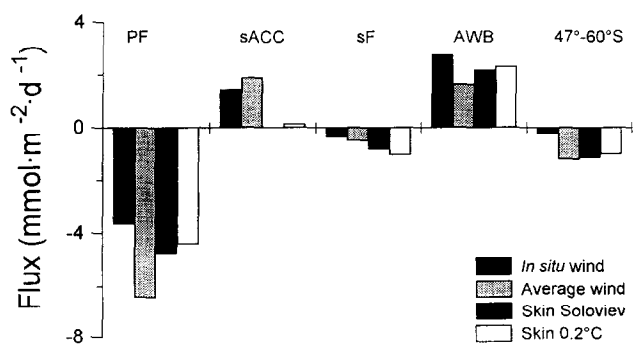


Fig. 12. Sea to air exchange of CO<sub>2</sub> for the hydrographic regions (Table 1) of ANT X/6 along 6°W from 11 October to 24 November 1992 with the appropriate Wanninkhof relationship (Equation (4a) and (4b), respectively) at *in situ* atmospheric pressure. Fluxes are shown for (i) *in situ* wind speed, no thermal skin effect; (ii) average wind speed, no thermal skin effect; (iii) *in situ* wind speed, skin effect by Soloviev and Schlüssel (1994a,b); and (iv) *in situ* wind speed, cold skin effect of 0.2°C. Positive fluxes indicate oceanic release of CO<sub>2</sub>. See also Table 3.

Table 4. Sea to air exchange of  $\text{CO}_2$  ( $\text{mmol m}^{-2} \text{ day}^{-1}$ ) at *in situ* and average wind speed with the appropriate Wanninkhof (1992) relationship (Equation 4(a) and 4(b), respectively) in the hydrographic regions (Table 1) along  $6^\circ\text{W}$  for individual transects of ANT X/6. Positive values indicate oceanic release of  $\text{CO}_2$ , negative fluxes denote oceanic uptake. See also Fig. 11

	PF	sACC	sF	AWB
<i>In situ wind speed</i>				
2	-0.9	2.5	-0.8	0.0
3	-3.6	0.0	-2.7	0.0
5	-5.2	0.9	-1.1	0.0
6	-1.0	0.8	1.4	7.2
7-10	—	—	—	4.0
11	-3.9	3.1	1.2	1.8
12	-7.4	—	—	—
X/6	-3.7	1.4	-0.4	2.7
<i>Average wind speed</i>				
2	-2.6	0.5	-0.7	0.0
3	-2.0	-0.4	-2.8	0.0
5	-4.7	2.1	-1.5	0.0
6	-3.7	4.5	1.5	3.0
7	—	—	—	2.6
11	-12.5	2.7	0.8	2.2
12	-13.1	—	—	—
X/6	-6.4	1.9	-0.5	1.6

the Wanninkhof relationship indicated 57% higher oceanic uptake of  $\text{CO}_2$  than the Liss-Merlivat relationship (Table 3). Summarising, the area between  $46.8^\circ$  and  $59.8^\circ\text{S}$  was a slight oceanic sink for  $\text{CO}_2$  of  $0.3 \text{ mmol m}^{-2} \text{ day}^{-1}$  at *in situ* wind speed with the Wanninkhof relationship (Table 3).

### Thermal skin effect

The sea surface skin layer is predicted to be generally colder than bulk waters. The correspondingly higher solubility  $K_0$  will enhance the influx of atmospheric  $\text{CO}_2$  into the ocean. The impact of a thermal skin effect on air-sea exchange was assessed for (i) an assumed uniform skin temperature difference of  $0.2^\circ\text{C}$ ; (ii) a skin temperature difference calculated from a physical model; and (iii) actual measurement of the skin temperature, the latter for transect 11 only.

Using a uniform cold skin effect of  $0.2^\circ\text{C}$  enhanced the influxes and decreased the effluxes of  $\text{CO}_2$ . In some instances an efflux would change into an influx (Fig. 12, Table 3). Throughout the cruise, the area between  $46.8^\circ$  and  $59.8^\circ\text{S}$  would become a three-fold stronger oceanic sink of  $1.0 \text{ mmol m}^{-2} \text{ day}^{-1}$  with a thermal skin effect (Table 3).

Modelled skin temperatures were  $0.0^\circ$ – $0.2^\circ\text{C}$  below the temperature of bulk water at 8 m depth for the parametrisations of Hasse (1971), Schlüssel *et al.* (1990), Saunders (1967), and Soloviev and Schlüssel (1994a,b) (Table 5, Fig. 13). As the models were developed for temperate waters, they are not necessarily applicable for the Southern Ocean. In particular, high wind speeds and low temperatures could affect skin temperature differences. Clearly models need to be developed and validated for the Southern Ocean.

Table 5. Mean values and standard deviations of detected and modelled skin temperature differences for transects 2–12 and transect 11 of ANT X/6. The detected skin temperature difference was corrected to an instrument angle of 40° with the vertical. For comparison detected values have also been corrected to an angle of 60° with the vertical. Skin temperature differences have been determined for models by Schlüssel *et al.* (1990, night-time values), Hasse (1971), Saunders (1967) and Soloviev and Schlüssel (1994a, 1994b). A positive skin temperature difference indicates a cold thermal skin effect. Sea to air exchange of CO<sub>2</sub> has been included for transect 11 at in situ wind speed and atmospheric pressure with the Wanninkhof relationship (Equation (4a)) without a skin effect, for a standard cold skin effect of 0.2°C, detected and modelled skin effects. Positive fluxes denote oceanic release of CO<sub>2</sub>

	$\Delta T$ , mean 2 to 12 (°C)	$\sigma_{n-1}$ 2 to 12 (°C)	$\Delta T$ , mean 11 (°C)	$\sigma_{n-1}$ 11 (°C)	Flux 11 (mmol m <sup>-2</sup> day <sup>-1</sup> )
<i>In situ</i> wind, $\Delta T = 0^\circ\text{C}$	—	—	—	—	0.2
$\Delta T = 0.2^\circ\text{C}$	—	—	—	—	-0.3
$\Delta T$ —detected (40°)	—	—	1.4	0.6	-3.2
$\Delta T$ —corrected to 60°	—	—	1.3	0.6	-3.2
Schlüssel <i>et al.</i>	0.0	0.2	0.0	0.1	0.3
Hasse	0.0	0.1	0.0	0.2	0.2
Saunders	0.1	0.1	0.1	0.1	0.0
Soloviev and Schlüssel	0.2	0.2	0.2	0.2	-0.1

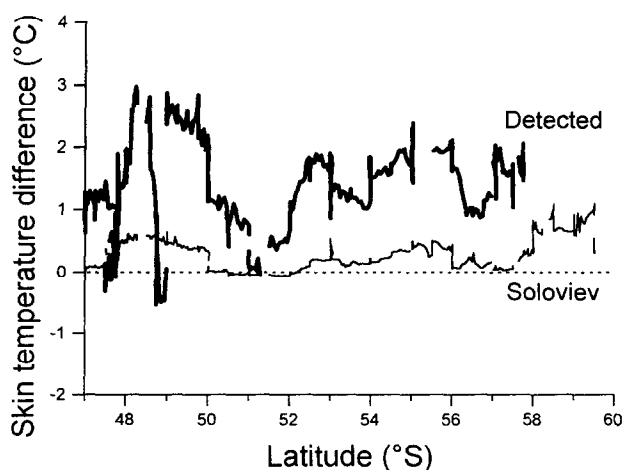


Fig. 13. Skin temperature differences detected and obtained with the model of Soloviev and Schlüssel (1994a,b) for transect 11 (Tables 2 and 5). Note that a positive skin temperature difference indicates a cold thermal skin effect.

Detected skin temperature was on average 1.4°C below that of bulk water with large variation (Fig. 13, Table 5). Correction to a range of instrument angles from 0 to 70° with the vertical did not affect the detected skin temperature difference largely. This was primarily, because the temperature of the cloud base, used as an estimate of the temperature of reflected radiation, only differed some degrees from the bulk water temperature. The detected skin temperature difference appears to follow the predicted trend of the model by Soloviev and Schlüssel (1994a,b), but is much larger and has a larger variation (Fig. 13). The discrepancy between detected and modelled skin temperatures could indicate that the outdoor measurements had an offset relative to the stable laboratory calibration. A possible

cause could be the effect of the instrument temperature on the detected signal, with calibration at room temperature and outdoor measurements at near-freezing conditions. Several other conceivable interferences (condensation of moisture on the foil covering the lens, an effect of the foil, emission of infrared radiation by water droplets in the path of the sensor) would be expected to give rise to more random offset, rather than the observed fairly systematic one (Fig. 13). Also, temperature gradients may exist in the 8 m interval between the registration of temperatures of bulk water and the skin layer. Finally, the reflection correction of the skin temperature measurements is not easily quantifiable.

For transect 11 application of the detected skin temperature differences considerably raised the oceanic uptake of  $\text{CO}_2$  relative to the small uptake with modelled skin temperatures (Table 5). Obviously suitable physical concepts and models as well as accurate measurement of the skin temperature in polar oceans and assessment of its effect on  $\text{CO}_2$ -exchange are not trivial matters.

### *CO<sub>2</sub>-signature of sea-ice brines*

Ice was young and contained low biological activity at the first ice station. At the second site, ice was melting fast, such that the sampled "sea-ice brine" was a mixture of brine, half-melted ice particles and debris. Brines were 0.5–0.7°C colder than surrounding seawater (Table 6). Sea-ice brines were more saline than seawater at the first site, while they were less saline at the second. The latter indicated mixing in of ice debris.

Sea-ice brines had very low nutrient contents and  $\text{fCO}_2$  as low as 100  $\mu\text{atm}$ . The  $\text{fCO}_2$ -values appeared to increase during (slow) sampling. The large undersaturation of  $\text{CO}_2$  and low nutrient contents of the brines were in accordance with Rutgers van der Loeff (1992) and Gleitz *et al.* (1995). These authors suggested that the chemical composition of brine in summer-ice was strongly influenced by primary production and other biological processes. Our samples of brine collected in spring supported this.

The internal consistency of the  $\text{CO}_2$ -chemistry of the sea-ice brines was poor. Detected DIC-values were 10% higher than values calculated from  $\text{fCO}_2$  and alkalinity. This may have been caused by non-ideal sampling conditions and air-sea exchange. Biogenic calcite particles, although not observed, if present, could have caused an overestimation of DIC and alkalinity in the unfiltered samples.

Table 6. *Properties of sea-ice brines and underlying seawater at 20 m depth, at the two ice stations*

	57.4°S, 6.2°W, 11 October 1992			59.3°S, 6.0°W, 6 November 1992		
	Seawater	Brine A	Brine B	Seawater	Brine C	Brine D
Temperature (°C)	−1.82	−2.3	−2.3	−1.83	−2.4	−2.5
Salinity	34.06	41.82	41.05	34.28	33.77	33.10
Alkalinity ( $\mu\text{eq kg}^{-1}$ )	2317	2595	2475	2339	2280	2216
DIC ( $\mu\text{mol kg}^{-1}$ )	2190	2451	2554	2203	—	—
$\text{fCO}_2$ ( $\mu\text{atm}$ )	—	115	115	—	81	77
Nitrate ( $\mu\text{mol kg}^{-1}$ )	29.1	0.4	0.0	27.6	0.3	0.1
Phosphate ( $\mu\text{mol kg}^{-1}$ )	2.01	0.3	0.5	2.04	0.2	0.2
Silicate ( $\mu\text{mol kg}^{-1}$ )	60.1	<9.1	<4.9	77.1	<2.6	<5.3
$\text{Alk}_{34}$ ( $\mu\text{eq kg}^{-1}$ )	2313	2109	2050	2320	2296	2276



Alkalinity values were normalized to a salinity of 34. Sea-ice brines at the first station had lower normalized alkalinity than seawater, possibly caused by precipitation of calcium carbonate ( $\text{CaCO}_3$ ) in the ice (Jones and Coote, 1981). The sequence of (i)  $\text{CaCO}_3$ -precipitation in the ice during freezing causing  $\text{CO}_2$ -supersaturation in the brine; (ii) leakage of brine; and (iii) partial  $\text{CaCO}_3$ -dissolution into the brine would result in the observed undersaturation of  $\text{CO}_2$  and relatively low normalized alkalinity. At the second station, normalised alkalinity of the brine was close to that of seawater, which would not suggest  $\text{CaCO}_3$ -precipitation.

The few studies available suggest strong undersaturation of  $\text{CO}_2$  and low nutrient contents of sea-ice brines in spring and summer due largely to biological activity and possibly to a sequence of  $\text{CaCO}_3$ -precipitation, brine leakage and  $\text{CaCO}_3$ -dissolution. However, the undersaturation of  $\text{CO}_2$  may not be universal as the variability of sea-ice extent and of its biological activity might well be reflected in variation of its  $\text{CO}_2$ -characteristics. The dynamic Antarctic sea-ice ecosystem, which accounts for 5–6% of the global oceans in winter and for 1–2% in summer (Hellmer and Bersch, 1985), may have a significant impact on global air–sea exchange of  $\text{CO}_2$ .

## DISCUSSION

### *Changes of $f\text{CO}_2$ in surface-water*

Observed daily changes of  $f\text{CO}_2$  in surface-water,  $\Delta f/\Delta t$ , were related to the expected changes of  $f\text{CO}_2$  by surface-water warming,  $(\delta f/\delta t)_T$ , and air–sea exchange,  $(\delta f/\delta t)_F$ , following the concepts of Poisson *et al.* (1993) (Table 7):

$$\Delta f/\Delta t = (\delta f/\delta t)_T + (\delta f/\delta t)_F + (\delta f/\delta t)_R. \quad (5)$$

The residual daily change of  $f\text{CO}_2$ ,  $(\delta f/\delta t)_R$ , was that part of the observed daily change, which was not explained by a temperature change and air–sea exchange. It was ascribed to the combined effects of biological activity, upwelling, mixing, variability of water masses flowing across  $6^\circ\text{W}$ , and ice melting. Neglecting the latter three physical processes, the residual daily change of  $f\text{CO}_2$  would be due to vertical advection by Ekman upwelling and biological activity.

The observed and expected changes of surface water  $f\text{CO}_2$  were estimated for the hydrographic regions for transects 3–6, 5–11, 6–11 and 3–11 (Table 7). Average values of surface-water  $f\text{CO}_2$  and temperature were calculated for each period and region from the online data. The daily thermodynamical change of  $f\text{CO}_2$ ,  $(\delta f/\delta t)_T$ , was calculated from the daily temperature change with the parametrization by Copin-Montégut (1988, 1989). The daily change of surface water  $f\text{CO}_2$  due air–sea exchange,  $(\delta f/\delta t)_F$ , was the product of average air–sea exchange  $F_{av}$ , the buffer factor  $\beta$ , and  $f\text{CO}_2$  divided by TDIC:

$$(\delta f/\delta t)_F = F_{av} \cdot \beta \cdot f\text{CO}_{2w} \cdot \text{TDIC}^{-1}. \quad (6)$$

The average air–sea flux of  $\text{CO}_2$ ,  $F_{av}$ , was taken as the average of the fluxes with the Wanninkhof relationship at actual wind speed over each period. The buffer factor or Revelle factor,  $\beta$ , related a relative change of  $f\text{CO}_2$  to the corresponding relative change of DIC (Poisson *et al.*, 1993):

$$\beta = \partial \ln(f\text{CO}_{2w}) \cdot \partial \ln(\text{DIC})^{-1}. \quad (7)$$

Table 7. Changes of surface-water  $f\text{CO}_2$  between transects 3, 5, 6 and 11 of ANT X/6 for the hydrographic regions between  $46.8^\circ$  and  $59.8^\circ\text{S}$  at  $6^\circ\text{W}$  (Table 1) in October–November 1992, following the concepts of Poisson et al. (1993). For the calculations we used the time interval  $\Delta t$ , the mixed layer depth, changes of surface-water temperature  $\Delta T$  and  $f\text{CO}_2$  ( $\Delta f\text{CO}_2$ ), values for surface water  $f\text{CO}_2$ , average dissolved inorganic carbon (DIC), and depth integrated total dissolved inorganic carbon in the mixed layer (TDIC) at the start of each period, and buffer factor  $\beta$ . The observed daily change of  $f\text{CO}_2$ ,  $\Delta f/\Delta t$ , equals the sum of thermodynamical change,  $(\delta f/\delta t)_T$ , the change by sea to air exchange,  $(\delta f/\delta t)_F$ , and the residual change  $(\delta f/\delta t)_R$  according to Equation (5):  $\Delta f/\Delta t = (\delta f/\delta t)_T + (\delta f/\delta t)_F + (\delta f/\delta t)_R$

	$\Delta t$ (d)	Depth ML (m)	$\Delta T$ ( $^\circ\text{C}$ )	$\Delta f\text{CO}_2$ ( $\mu\text{atm}$ )	$f\text{CO}_{2w}$ ( $\mu\text{atm}$ )	DIC ( $\mu\text{mol kg}^{-1}$ )	TDIC ( $\text{mol m}^{-2}$ )	Buffer factor	$F_{av}$ ( $\text{mmol m}^{-2}$ $\text{day}^{-1}$ )	$\Delta f/\Delta t$ ( $\mu\text{atm day}^{-1}$ )	$(\delta f/\delta t)_T$ ( $\mu\text{atm day}^{-1}$ )	$(\delta f/\delta t)_F$ ( $\mu\text{atm day}^{-1}$ )	$(\delta f/\delta t)_R$ ( $\mu\text{atm day}^{-1}$ )
3-6	PF 12.5	80	0.47	-6.2	342.4	2130	175	14.9	-3.2	-0.49	0.52	0.09	-1.11
	sACC 12.0	100	0.24	6.4	349.6	2155	221	15.1	0.6	0.53	0.30	-0.01	0.25
	sF 12.1	80	0.31	3.7	345.2	2157	177	15.4	-0.8	0.31	0.39	0.02	-0.11
5-11	PF 20.8	80	0.69	-17.5	339.0	2125	175	14.2	-3.4	-0.84	0.47	0.09	-1.40
	sACC 20.3	100	0.30	1.8	355.9	2146	220	15.1	1.6	0.09	0.23	-0.04	-0.10
	sF 20.1	80	0.45	5.4	347.3	2161	178	15.4	0.5	0.27	0.32	-0.02	-0.04
6-11	PF 18.6	80	0.68	-14.7	336.2	2124	174	14.2	-2.5	-0.79	0.53	0.07	-1.39
	sACC 14.5	100	0.25	1.8	355.9	2152	221	15.1	2.0	0.12	0.27	-0.05	-0.09
	sF 11.3	80	0.15	3.7	349.0	2145	176	15.4	1.3	0.33	0.19	-0.04	0.18
	AWB 7.5	80	0.06	-1.1	362.6	2185	180	15.4	4.3	-0.15	0.12	-0.13	-0.13
3-11	PF 31.2	80	1.15	-20.9	342.4	2130	175	14.9	-3.4	-0.67	0.54	0.10	-1.31
	sACC 26.6	100	0.48	8.2	349.6	2155	221	15.1	1.2	0.31	0.28	-0.03	0.06
	sF 23.4	80	0.46	7.4	345.2	2157	177	15.4	-0.3	0.32	0.30	0.01	0.01

The average DIC in the mixed layer was estimated from DIC detected in CTD-samples. Mixed layer depths were inferred from potential density profiles of CTD-casts. The total amount of dissolved inorganic carbon in the mixed layer, TDIC, was estimated as the product of average DIC in the mixed layer, the mixed layer depth, and the density for the start of each period. To determine the buffer factor, alkalinity was calculated for  $f\text{CO}_2$  of 350  $\mu\text{atm}$  and detected DIC with dissociation constants of Goyet and Poisson (1989). Then DIC was lowered 10  $\mu\text{mol kg}^{-1}$  at constant alkalinity to find the corresponding change of  $f\text{CO}_2$ .

*Polar Frontal region.* In the Polar Frontal region  $f\text{CO}_2$  of surface-water decreased at 0.67  $\mu\text{atm day}^{-1}$  between transects 3 and 11 (Table 7). This change would have been almost twice as large if a rise of water temperature with 1.2°C had not counteracted the decrease with 0.54  $\mu\text{atm day}^{-1}$ . Air-sea exchange supplied  $\text{CO}_2$  at a rate of 0.10  $\mu\text{atm day}^{-1}$ . Most of the negative residual daily change of 1.31  $\mu\text{atm day}^{-1}$  corresponded probably to biological uptake of dissolved  $\text{CO}_2$  for photosynthesis as suggested by the increases of primary production (Jochem *et al.*, 1995) and chlorophyll *a* (Fig. 7). The occurrence of phytoplankton blooms was largely dominated by iron being available for phytoplankton growth, but other factors such as light limitation and grazing also played an important role (de Baar *et al.*, 1995; Sullivan *et al.*, 1993).

*Southern ACC and southern Frontal region.* In the sACC and the sF a surface-water warming of 0.5°C was responsible for most of the observed increases of  $f\text{CO}_2$  with 0.31 and 0.32  $\mu\text{atm day}^{-1}$  throughout the cruise (Table 7). Air-sea exchange only had a small influence. Residual daily changes of  $f\text{CO}_2$  were negligible between transects 3 and 11. Moderate residual daily changes were observed between transects 3 and 6 and between 6 and 11. These could have been caused by heterogeneity of water reaching the area or by ice melting in case of the sF between transects 3 and 6.

*ACC–Weddell Gyre boundary.* For the AWB region only 8 days passed between transects 6 and 11. Therefore, not much weight should be attached to the moderate observed and residual daily changes of surface water  $f\text{CO}_2$  (Table 7). Water below the ice might well have been heterogeneous, and ice melting could have affected  $f\text{CO}_2$ . In addition, the frontal jet of the AWB could have contained large meanders.

#### *Changes of dissolved inorganic carbon in the mixed layer*

The residual daily changes of  $f\text{CO}_2$ ,  $(\delta f/\delta t)_R$ , were converted to a net residual daily change of dissolved inorganic carbon over the depth of the mixed layer,  $R_{f\text{CO}_2}$ :

$$R_{f\text{CO}_2} = (\delta f/\delta t)_R \cdot \text{TDIC} \cdot \beta^{-1} \cdot f\text{CO}_{2w}^{-1}. \quad (8)$$

Alternatively, the term  $R_{\text{DIC}}$  was calculated from the measured changes of DIC, where the latter  $R_{\text{DIC}}$  had been corrected for the air-sea exchange flux. As mentioned above, the residual change,  $R$ , corresponded to the effects of vertical advection,  $U$ , and biological activity. The biological fixation of inorganic carbon could be quantified by net primary production  $P$ , minus mineralization  $M$ , leading to biomass increase  $B$  and some export sedimentation  $E$ . Hence:

$$\Delta \text{TDIC} = \text{Upward transport} - \text{Primary production} + \text{Mineralization}$$

$$= \text{Upward transport} - \text{Biomass increase} - \text{Export}$$

$$R = U - P + M = U - B - E. \quad (9)$$

Residual daily changes of the dissolved inorganic carbon content in the mixed layer ( $R_{\text{fCO}_2}$  and  $R_{\text{DIC}}$ ), vertical advection  $U$ , net primary production  $P$  (after Jochem *et al.*, 1995) and biomass increase  $B$  (after Bathmann *et al.*, 1997) have been calculated between transects (Table 8). No direct measurements were made of mineralization  $M$  and carbon export  $E$ . Rather  $M$  and  $E$  were inferred using Equation (9). In the Polar Frontal region the calculated export  $E$  between transects 5 and 11 could be compared with a value based on the  $^{234}\text{Th}$ -deficiency method (Rutgers van der Loeff *et al.*, 1997).

From Table 7 values were adopted for the time interval, mixed layer depth,  $\text{fCO}_2$ , DIC, TDIC, buffer factor, air-sea exchange and the residual daily changes of  $\text{fCO}_2$ ,  $(\delta f/\delta t)_{\text{R}}$ . Residual daily changes of dissolved inorganic carbon over the depth of the mixed layer on the basis of  $\text{fCO}_2$  ( $R_{\text{fCO}_2}$ ) were derived by multiplying residual daily changes of  $\text{fCO}_2$ ,  $(\delta f/\delta t)_{\text{R}}$ , with TDIC, and by dividing this product by the buffer factor and  $\text{fCO}_2$  (Table 7). The second series of values for residual daily changes of dissolved inorganic carbon ( $R_{\text{DIC}}$ ) was obtained by correcting observed changes of DIC in the mixed layer for the air-sea exchange flux  $F$ .

An estimate of the upward flux of inorganic carbon by Ekman pumping,  $U$ , has been derived analogous to the procedure in de Baar *et al.* (1995):

$$U = -K_z \cdot (\delta \text{DIC}/\delta z). \quad (10)$$

The value for the diffusivity  $K_z$  of  $3 \cdot 10^{-5} \text{ m}^2 \cdot \text{s}^{-1}$  was taken from de Baar *et al.* (1995). The increase of DIC of  $20\text{--}50 \mu\text{mol kg}^{-1}$  between 100 and 200 m depth for transects 5 and 11 corresponded to upward fluxes of dissolved inorganic carbon of  $0.8 \text{ mmol m}^{-2} \text{ day}^{-1}$  in the PF and of  $1.0 \text{ mmol m}^{-2} \text{ day}^{-1}$  in the sACC, sF and AWB (Table 8).

Values of  $P$  from  $^{14}\text{C}$  primary production (Jochem *et al.*, 1995) and  $B$  from POC change (Bathmann *et al.*, 1997) in the mixed layer were obtained from the CTD-database for the hydrographic regions of the transects. This procedure, rather than applying the estimates indicated in the individual papers (Jochem *et al.*, 1995; Bathmann *et al.*, 1997), ensured that values had been derived for the same depths and time intervals and as such were comparable. It was assumed that changes of the dissolved organic carbon (DOC) content were not significant, judging from surface water DOC-values being rather uniform with time, albeit somewhat higher than in deeper waters (Kähler *et al.*, 1997). Therefore the estimate  $B$  of biomass growth was estimated solely from changes in particulate organic carbon (POC).

$R_{\text{fCO}_2}$  and  $R_{\text{DIC}}$ . Residual daily changes of dissolved inorganic carbon based on DIC ( $R_{\text{DIC}}$ ) indicated more variability than those based on  $\text{fCO}_2$  ( $R_{\text{fCO}_2}$ ). The latter residual daily changes showed a consistent pattern of losses of dissolved inorganic carbon in the Polar Frontal region and low to moderate changes further south. The difference between both estimates of the residual daily change  $R$  may be explained from the high sampling density of  $\text{fCO}_2$  compared to the lower density of DIC samples from CTD-casts. In addition, planned CTD-casts could not always be executed in case of storms, while online surface water

Table 8. Estimates of the various terms of Equation (9):  $R = U - P + M = U - B - E$  between transects 3, 5, 6 and 11 for the hydrographic regions along 6°W (Table 1). The residual daily change  $R_{DIC}$  of the dissolved inorganic carbon content of the mixed layer was derived from measured changes of DIC  $R_{\Delta DIC}$  after correction for air-sea exchange (see text). Alternatively, the residual daily change  $R_{CO_2}$  of DIC in the mixed layer was estimated from daily changes  $(\delta f(\delta t)_R$  of  $fCO_2$  (Table 7) and the conversion  $R_{fCO_2} = (\delta f(\delta t)_R / TDIC) \cdot \beta^{-1} \cdot fCO_{2w}^{-1}$  (Equation (8)). Upward transport  $U$ , net primary production  $P$  (Jochem *et al.*, 1995), and biomass growth  $B$  from POC (Bathmann *et al.*, 1997) were derived as in the text. Apparent mineralization  $M$  and carbon export  $E$  were estimated by difference according to:  $R = U - P + M = U - B - E$  (Equation (9)), with  $R$  relying on  $R_{fCO_2}$  rather than  $R_{DIC}$ .

		$^{14}C$ prod ( $mg\ m^{-2}$ $day^{-1}$ )	dPOC ( $mmol\ m^{-3}$ )	dDIC ( $\mu mol\ kg^{-1}$ )	$R_{\Delta DIC}$ ( $mmol\ m^{-2}$ $day^{-1}$ )	$R_{DIC}$ ( $mmol\ m^{-2}$ $day^{-1}$ )	$R_{CO_2}$ ( $mmol\ m^{-2}$ $day^{-1}$ )	$U$ (mmol $m^{-2}$ $day^{-1}$ )	$P$ (mmol $m^{-2}$ $day^{-1}$ )	$M$ (mmol $m^{-2}$ $day^{-1}$ )	$B$ (mmol $m^{-2}$ $day^{-1}$ )	$E$ (mmol $m^{-2}$ $day^{-1}$ )
3-6	PF	571	—	—	—	—	-38.0	0.8	40.8	2.0	—	—
	sACC	247	—	—	—	—	10.3	1.0	17.6	26.9	—	—
	sF	202	—	-12.6	-85.6	-86.4	-3.6	1.0	14.4	9.8	—	—
5-11	PF	1362	8.3	-2.0	-8.1	-11.4	-51.0	0.8	97.3	45.5	31.8	20.0
	sACC	236	3.0	12.1	61.1	62.7	-4.2	1.0	16.9	11.6	14.9	(-9.6)
	sF	249	—	6.6	26.8	27.3	-1.3	1.0	17.8	15.5	—	—
6-11	PF	1362	—	—	—	—	-50.9	0.8	97.3	45.6	—	—
	sACC	236	—	—	—	—	-3.9	1.0	16.9	12.0	—	—
	sF	249	—	23.0	167.1	168.4	6.1	1.0	17.8	22.8	—	—
3-11	AWB	283	-0.9	-1.0	-10.4	-6.1	-4.2	1.0	20.2	15.0	-9.8	15.0
	PF	1039	11.6	-7.2	-18.9	-22.3	-44.8	0.8	74.2	28.7	29.7	15.9
	sACC	236	—	—	—	—	2.3	1.0	16.9	18.2	—	—
3-11	sF	227	—	10.4	36.4	36.1	0.3	1.0	16.2	15.4	—	—

sampling continued. Hence, a latitudinal gradient of DIC within the hydrographic regions may have contributed to the residual daily changes based on DIC,  $R_{\text{DIC}}$ .

*Vertical advection.* Values for vertical advection of inorganic carbon were small in all four regions (Table 8). In the sACC, sF and AWB, the advection term  $U$  had virtually the same magnitude as the residual daily changes of inorganic carbon  $R_{\text{fCO}_2}$ . In the Polar Front the daily residual change was much larger, see below. In all four regions vertical advection was low in comparison to primary production and changes in biomass.

*Polar Frontal region.* In the Polar Frontal region, the average primary production of  $40.8 \text{ mmol m}^{-2} \text{ day}^{-1}$  between transects 3 and 6 doubled to  $97.3 \text{ mmol m}^{-2} \text{ day}^{-1}$  between transects 6 and 11 (Table 8). Low apparent mineralization of  $2.0 \text{ mmol m}^{-2} \text{ day}^{-1}$  between transects 3 and 6 had become  $45.6 \text{ mmol m}^{-2} \text{ day}^{-1}$ , or 47% of primary production between transects 6 and 11. The increase of biomass was close to  $30 \text{ mmol m}^{-2} \text{ day}^{-1}$  throughout the cruise. Between transects 5 and 11 the apparent carbon export was  $20.0 \text{ mmol m}^{-2} \text{ day}^{-1}$ , or 21% of primary production.

From  $^{234}\text{Th}$ -data carbon export was estimated as  $21\text{--}41 \text{ mmol m}^{-2} \text{ day}^{-1}$  ( $0.43\text{--}0.86 \text{ mol m}^{-2}$  over 20.8 days) between transects 5 and 11 for  $47\text{--}48^\circ\text{S}$  (Rutgers van der Loeff *et al.*, 1997). This corresponded to 21–42% of primary production in the Polar Frontal region. Our value of  $20.0 \text{ mmol m}^{-2} \text{ day}^{-1}$  for carbon export is close to the lower limit of the range given by Rutgers van der Loeff *et al.* (1997)

*Southern ACC and southern Frontal region.* In the southern ACC and southern Frontal region residual daily changes of dissolved inorganic carbon based on  $\text{fCO}_2$ ,  $R_{\text{fCO}_2}$ , were much smaller and less significant than those in the Polar Frontal region. Primary production was rather stable at  $14.4\text{--}17.8 \text{ mmol m}^{-2} \text{ day}^{-1}$  throughout the cruise (Table 8). Estimates of mineralization varied from  $9.8\text{--}26.9 \text{ mmol m}^{-2} \text{ day}^{-1}$ . This suggested an efficient cycling of carbon in the mixed layer.

*ACC–Weddell Gyre boundary.* In the AWB the low residual daily change  $R_{\text{fCO}_2}$  and primary production of  $20.2 \text{ mmol m}^{-2} \text{ day}^{-1}$  suggested an efficient cycling of carbon within the mixed layer as for the sACC and sF. The observed decrease of POC would indicate a surprisingly high carbon export.

### Seasonal $\text{fCO}_2$ -changes

Having assessed the mechanisms behind changes of surface water  $\text{fCO}_2$  in austral spring 1992, a comparison with other seasons and years is of interest. In January through April in the years 1984–1990 the average  $\text{CO}_2$ -content of surface water was slightly undersaturated between  $47^\circ\text{S}$  and  $56^\circ\text{S}$  along  $6^\circ\text{W}$  and moderately undersaturated further south (Takahashi *et al.*, 1993). Apparently some months further in the season the biological uptake results in a more general undersaturation of  $\text{CO}_2$ .

*Polar Frontal region.* In spring 1992 the average  $\Delta\text{fCO}_2$  between surface water and air decreased from  $-10$  to  $-35 \text{ } \mu\text{atm}$  from transect 2 to 11 in the Polar Frontal region, while in summer 1984–1990, the average  $\text{pCO}_2$  was  $0$  to  $-20 \text{ } \mu\text{atm}$  (Takahashi *et al.*, 1993). Less intense biological productivity in summer than in spring, possibly due to grazing or iron

depletion (de Baar *et al.*, 1995), could explain the larger undersaturation of surface water in spring than in summer.

*Southern ACC and southern Frontal region.* Seasonal warming changed the sACC and the sF from a slight sink in October to a small source in November 1992. In summer, the  $\Delta p\text{CO}_2$  of surface water was 0 to  $-20 \mu\text{atm}$  (Takahashi *et al.*, 1993). The change from source in November to sink in summer could be the result of biological activity and a lower rate of surface water warming in late spring and summer.

*ACC–Weddell Gyre Boundary and Weddell Gyre.* In the mostly ice-covered Boundary region and Weddell Gyre proper ( $56$ – $60^\circ\text{S}$ ) surface water was supersaturated with  $\text{CO}_2$  and undersaturated with  $\text{O}_2$  in November 1992. Winter measurements in August–September 1986 also showed supersaturated  $\text{CO}_2$ -values (Chipman, Keeling and Weiss in Schnack-Schiel, 1987; Weiss *et al.*, 1992) and undersaturated  $\text{O}_2$ -concentrations (Gordon and Huber, 1990) in the mixed layer of the ice-covered Weddell Sea. Gordon and Huber (1990) argued that upwelling of Warm Deep Water and not *in situ* biological processes had caused the undersaturation of  $\text{O}_2$ . These undersaturated  $\text{O}_2$ -values in late winter 1986 agreed well with  $50 \mu\text{mol kg}^{-1}$  undersaturated  $\text{O}_2$ -values in remnant Winter Water (Weiss *et al.*, 1979) and with an  $\text{O}_2$ -saturation of 80–90% in surface water in October–November 1988 (Bouquegneau *et al.*, 1992) and November 1992 (this study).

In summer the  $p\text{CO}_2$  of surface waters of the Weddell Sea were 20 to over  $40 \mu\text{atm}$  below the atmospheric value in January through April in the years 1984–1990 (Takahashi *et al.*, 1993). A low  $\Delta p\text{CO}_2$  of  $-30$  to  $-200 \mu\text{atm}$  was calculated for surface water of the central Weddell Sea in December–January 1992–1993 (Hoppema *et al.*, 1995). Also summer surface water had a lower and more variable specific DIC content (DIC divided by salinity) than remnant Winter Water (Weiss *et al.*, 1979). Concentrations of  $\text{O}_2$  in summer surface water varied in a broad range around equilibrium solubility and were well above the undersaturated values in remnant Winter Water (Weiss *et al.*, 1979). This indicated air–sea exchange and considerable biological activity after the disappearance of the ice-cover in late spring (Weiss *et al.*, 1979).

It appears that, in the winter, surface water of the Weddell Sea gradually becomes oversaturated with  $\text{CO}_2$  and undersaturated with  $\text{O}_2$  due to continuous upwelling of  $\text{CO}_2$ -rich and  $\text{O}_2$ -poor Warm Deep Water, while the ice-cover inhibits air–sea gas exchange. The low  $\Delta p\text{CO}_2$  of  $-20$  to  $-50 \mu\text{atm}$  in surface water of the ice-covered Weddell Sea at the onset of winter in June–July 1992 (Hoppema *et al.*, 1995) may represent the transition between strong undersaturation of  $\text{CO}_2$  in summer and oversaturation below the ice in (late) winter and early spring. High biological activity lowers  $f\text{CO}_2$  and increases  $\text{O}_2$ -levels in summer. Thus, the Weddell Gyre acts as a  $\text{CO}_2$ -source immediately after retreat of the ice-cover, but changes into a  $\text{CO}_2$ -sink in late spring or early summer. As yet, it is impossible to identify from surface water  $f\text{CO}_2$ -data whether the Weddell Gyre is a net oceanic source or sink of  $\text{CO}_2$  at an annual basis.

#### *Air–sea gas exchange*

Published estimates of  $\text{CO}_2$ -fluxes depend on a wide range of selected wind intervals (Table 9. Data sources: <sup>a</sup>Tourre *et al.*, 1986; <sup>b</sup>Bacastow and Maier-Reimer, 1990; <sup>c</sup>Broecker *et al.*, 1986; <sup>d</sup>Takahashi *et al.*, 1986, 1991; <sup>e</sup>Japan Meteorological Agency; <sup>f</sup>Norwegian

Table 9. The parametrization of air-sea exchange, origin and time intervals of wind speed, and of surface-water  $f\text{CO}_2$  ( $p\text{CO}_2$ ) for reported air-sea fluxes of  $\text{CO}_2$ . Numbers and abbreviations refer to studies, methods and data sources mentioned below. The level of extrapolation of the estimated fluxes differs largely

Study*	Method	Wind speed: source and interval	Surface-water $f\text{CO}_2$ : source and interval	Period	Temporal and spatial scale of flux estimate
1	LM	1 month, $1.9^\circ\text{--}1.9^\circ$ <sup>a</sup>	CTDs	07/82–08/84	Annual, regional
2	Rn	$z = 23\text{--}50\ \mu\text{m}$	pH, alkalinity	01–03/79	Period, regional
3	T	Ship	CTD, daily	26/04–07/05/89	Period, <i>in situ</i>
4	LM	$u = 5\ \text{m s}^{-1}$	Ship	—	Period, regional
5	LM	Satellite, 1 month, $2.5^\circ\text{--}2.5^\circ$	global $\Delta p\text{CO}_2$ <sup>b,c</sup>	—	Annual, global
6	LM	Satellite, 3 months, $10^\circ\text{--}10^\circ$	global $\Delta p\text{CO}_2$ <sup>c</sup>	—	Annual, global
7	U	Ship, 1 day	Online, 20 min	04/83, 03–04/84	Period, regional
8	W	1 month <sup>d</sup>	Online, 1 h	1991–92	Period, regional
9	LM		DIC, salinity	07–08/83	Period, regional
10	LM, T	12 h, $2.5^\circ\text{--}2.5^\circ$ <sup>e</sup>	Online, 1 h	01/87–02/89	Period, regional
11	Z	$z = 40\ \mu\text{m}$	CTDs	05–06/86	Period, regional
12	LM	6 h, $75\text{--}75\ \text{km}^2$ <sup>f</sup>	Online, 30 min	07–08/81, 02–04/82, 08–09/91	Annual, regional
13	LM, W	Ship, 1 h	Online, 10 min	01–09/91, 92, 93	Annual, regional
14	LM, T	30 day, 90 day; ship, 12 h <sup>g,h</sup>	Online, 1 h mean of 30 min value	Autumns 1984–89	Period, regional
15	LM	Ship, 1 day	CTDs	01–03/93	Period, regional
16	LM	Ship, 712 h	CTDs	—	Period, <i>in situ</i>
17	E	$E = 20\ \text{mol m}^{-2}\ \text{y}^{-1}$	CTDs	03/83–05/85	Annual, <i>in situ</i>
18	LM	Ship, 2 day	—	01–09/91	Period, regional
19	LM, T, W	Ship, 1 h	Online, 6 min	11/92–03/93	Period, regional
20	LM, W	Ship, 1 h	Online, 7 min	05–06/89	Period, <i>in situ</i>
21	K	1 y, $10^\circ\text{--}10^\circ$ <sup>i</sup>	CTDs	04–10/1903–73	Annual, regional
22	LM	Ship's mean $u = 10\text{--}15\ \text{m s}^{-1}$	Online, 3 min	05–06/91	Period, regional
23	Rn	—	CTDs	11/82–02/83	Period, regional
24	T	1 month <sup>j</sup>	Online	1984–89	—
25	T	1 month, $2^\circ\text{--}2^\circ$ <sup>k</sup>	CTDs; online, 2° data <sup>k</sup>	1972–89	Annual, global
26	T	Buoy, 1 y, $u = 7\ \text{m s}^{-1}$	DIC, alkalinity, 1 month	10/88–12/92	Annual, regional
27	LM	Ship, 3 h	Online, 30 min	08/73–07/78	Period, <i>in situ</i>
28	LM	ship, $1^\circ$ <sup>l</sup>	Online	01/85–02/88	Period, regional
29	LM	Ship, 1 month	DIC, alkalinity	07–08/92	Period, regional

\*Studies: <sup>1</sup>Andrié *et al.*, 1986; <sup>2</sup>Batrakov *et al.*, 1981; <sup>3</sup>Chipman *et al.*, 1993; <sup>4</sup>Copin-Montégut, 1993; <sup>5</sup>Etcheto *et al.*, 1991; <sup>6</sup>Etcheto and Merlivat, 1988; <sup>7</sup>Feely *et al.*, 1987; <sup>8</sup>Feely *et al.*, 1995; <sup>9</sup>Garçon *et al.*, 1989; <sup>10</sup>Inoue and Sugimura, 1992; <sup>11</sup>Kempe and Pegler, 1991; <sup>12</sup>Lundberg, 1994; <sup>13</sup>Metzl *et al.*, 1995; <sup>14</sup>Murphy *et al.*, 1991a; <sup>15</sup>Oudot *et al.*, 1995; <sup>16</sup>Oudot and Andrié, 1989; <sup>17</sup>Peng *et al.*, 1987; <sup>18</sup>Poisson *et al.*, 1993; <sup>19</sup>Robertson and Watson, 1995; <sup>20</sup>Robertson *et al.*, 1993; <sup>21</sup>Roos and Gravenhorst, 1984; <sup>22</sup>Schneider *et al.*, 1992; <sup>23</sup>Smethie *et al.*, 1985; <sup>24</sup>Stephens *et al.*, 1995; <sup>25</sup>Tans *et al.*, 1990; <sup>26</sup>Winn *et al.*, 1995; <sup>27</sup>Wong and Chan, 1991; <sup>28</sup>Wong *et al.*, 1993, 1995; <sup>29</sup>Yager *et al.*, 1995.

Methods: E (Broecker and Peng, 1982),  $F = E \cdot \Delta p\text{CO}_2 / (p\text{CO}_{2\text{air}})$ ; K (Kromer, 1979),  $k_{\text{Rn}} = 0.72 \cdot (u - 4) + 0.3$ ,  $u > 4\ \text{m s}^{-1}$ ; LM (Liss and Merlivat, 1986); Rn (Smethie *et al.*, 1985),  $k_{\text{Rn}}$  from *in situ* Rn-profiles; T (Tans *et al.*, 1990),  $F = 0.016 \cdot (u - 3)$ ,  $u > 3\ \text{m s}^{-1}$ ; U (Smethie *et al.*, 1985)  $k_{\text{Rn}} = 0.28 \cdot u - 3.9$ ,  $u = 24\ \text{h}$  shipboard wind speed; W (Wanninkhof, 1992); Z (Broecker and Peng, 1982),  $F = D \cdot \Delta[\text{CO}_2] \cdot z^{-1}$ .

Data sources: <sup>a</sup>Tourre *et al.*, 1986; <sup>b</sup>Bacastow and Maier-Reimer, 1990; <sup>c</sup>Broecker *et al.*, 1986; <sup>d</sup>Takahashi *et al.*, 1986; Takahashi *et al.*, 1991; <sup>e</sup>Japan Meteorological Agency; <sup>f</sup>Norwegian Meteorological Institute; <sup>g</sup>Harrison, 1989; National Climate Centre 30 year climatology; Navy Fleet Numerical Oceanography Center; <sup>h</sup>Esbensen and Kushnir, 1981, Climatic Research Institute 150-year climatology; <sup>i</sup>U.S. Navy, 1974; <sup>j</sup>European Centre for Medium Range Weather Forecasts; <sup>k</sup>Lamont Doherty Geological Observatory; Roos and Gravenhorst, 1984; Andrié *et al.*, 1986; Gammon; <sup>l</sup>National Climatic Data Center, U.S.A.

Meteorological Institute; <sup>8</sup>Harrison, 1989; National Climate Centre 30-year climatology; Navy Fleet Numerical Oceanography Center; <sup>h</sup>Esbensen and Kushnir, 1981, Climatic Research Institute 150-year climatology; <sup>i</sup>U.S. Navy, 1974; <sup>j</sup>European Centre for Medium Range Weather Forecasts; <sup>k</sup>Lamont Doherty Geological Observatory; Roos and Gravenhorst, 1984; Andrié *et al.*, 1986; Gammon; <sup>l</sup>National Climatic Data Center, USA). The use of a short-term wind speed is optimal for the calculation of *in situ* air-sea exchange



fluxes. For the calculation of long-term fluxes one should consider that both its driving force, the concentration difference, and the transfer velocity change over time scales of hours and days, respectively. The ideal long-term estimate of the air–sea exchange flux is the average of an infinite number of *in situ* determined fluxes. Estimates, which are averages of several *in situ* fluxes, may come nearest to this. Even then, it remains a question as to whether it is better to use *in situ* wind speed for each individual flux or a wind speed with an appropriately short averaging interval. Our data demonstrate how much the estimated exchange for the area throughout the cruise is affected by the choice of *in situ* vs average wind speed.

Use of the realistic skin temperature differences considerably changed small air–sea exchange of CO<sub>2</sub> (Tables 3 and 5). Accurate measurement of the skin temperature and assessment of its effect on air–sea exchange of CO<sub>2</sub> need further study, and are crucial for any seasonal, regional or global assessment of CO<sub>2</sub>-fluxes between the atmosphere and the ocean.

### SUMMARY

The above demonstrates the complexities of variable fCO<sub>2</sub> in surface waters of an extensive region. The most promising approach to obtain as much information as possible from cruises is to study fCO<sub>2</sub> in relation to past and present biological, chemical and physical processes, as was demonstrated earlier by many researchers (Brewer, 1986; Poisson *et al.*, 1993; Takahashi *et al.*, 1993; Watson *et al.*, 1991; among others). Here the combined observations allowed detection of changes in the relationship between fCO<sub>2</sub> and chlorophyll *a*, and a first attempt to provide a budget of dissolved inorganic carbon in the mixed layer of the region. Hydrographic regions often differ in the processes affecting surface water fCO<sub>2</sub>. Hence, they should be treated separately when studying the mechanisms behind changes of surface water fCO<sub>2</sub>.

A general feature appears to be undersaturation of the surface water CO<sub>2</sub>-content in the Polar Frontal region. Seasonal bloom development causes large undersaturation, albeit counteracted by surface water warming. Surface waters of the sACC and the sF become supersaturated with CO<sub>2</sub> in spring due to seasonal warming. The Boundary region and the Weddell Gyre change from a CO<sub>2</sub>-source after retreat of the ice edge in spring to a sink later in the season, most likely due to strong bloom development.

The area between 46.8° and 59.8°S at 6°W was a sink for atmospheric CO<sub>2</sub> of 0.3 mmol m<sup>-2</sup> day<sup>-1</sup> at *in situ* wind speed from 11 October to 24 November 1992. The length of the wind speed interval and the actual skin effect deserve attention, when estimating long-term air–sea fluxes. Future studies of the gas transfer velocity need to address the effect of the skin layer on air–sea gas exchange directly. Large seasonal and spatial variability of the air–sea flux precludes a reliable estimate of the basin-wide flux for the Southern Ocean.

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